

Ecological and Health Assessment of Heavy Metals Present in the Soil of E-Waste Collection, Recycling and Dismantling Sites in the Region of DAKAR, Senegal

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Abstract

Soil pollution from waste electrical and electronic equipment is a growing concern in Africa. The heavy metals contained in this waste are now one of the main sources of pollution and the handling of waste electrical and electronic equipment (WEEE) is increasingly a health and environmental concern. In Dakar, several handling points of this equipment were identified and our study investigated the presence of trace metals in the soil of e-waste handling sites in the Dakar region and about the spatial distribution of Pb, Cd, Cu, Cr, Fe, As, Hg and Mn according to the activity carried out on these sites. Three categories of activity were identified, namely collection, dismantling and recycling and a control site.

The results show an abundance of metals for the collection and recycling sites which were such as Fe> Mn> Pb> Cu> Cr> Cd> As> Hg. In contrast, at the dismantling sites, the average total concentration of these elements was classified as Fe> Pb> Mn> Cu> Cr> As> Cd> Hg and that of the control site was Fe> Mn> Cu> Pb> Cr> Cd> As> Hg. The estimation of the potential ecological risk (Er) in relation to the control site shows an average risk coefficient for Fe, Mn and Cr below 40 for all sites, as well as for As in the collection and recycling sites and for Cu in the collection sites. Pb, Cd and Hg showed an average level of ecological "very high risk" for all sites and As also showed a "very high risk" for the dismantling sites for an ecological risk index of "Very high ecological risk" (RI> 600) for all sites associated with a non-carcinogenic risk greater than 1 for all site categories.

Keywords: Environmental Concern; Technological revolutions; Recycling

Abbreviations: WEEE: Waste Electrical and Electronic Equipment, CDI: Chronic Daily Intake, IARC: International Cancer Research Agency, TCR: Total Carcinogenic Risk.

Introduction

Electrical and electronic devices are increasingly playing an important role in the economic, industrial and social

activities of our society. Due to the various technological revolutions that have taken place around the world in recent years, these devices continue to evolve every day, which favors the creation of new devices even more efficient [1,2]. In order to meet the different economic and social requirements of society, the selling price of these devices may decrease depending on the country, which is sometimes to the detriment of the quality and lifespan of the equipment. This phenomenon is all the more responsible for the rapid generation of a new form of waste in the world called waste electrical and electronic equipment (WEEE or e-waste) [3,1].

E-waste has existed since the U.S. Resource Conservation and Recovery Act [4,5], and its sound management faces several challenges, including the alarming increase in the number of wastes, the way they are treated in developing countries and the impact they may have on health and the environment [5].

Waste is part of the fast-growing category of waste worldwide and its collection, recycling and dismantling system plays a key role in the management policy and effectiveness of this system [2,5,6,]. In Africa, the collection, recycling and dismantling of e-waste is usually done informally and sometimes in high-density areas. Some developing countries, including Ghana, Nigeria and Tanzania, are taking advantage of this practice by collecting and selling electronic equipment components [7]. This same practice is also observed in Senegal where recycling and dismantling of electronic waste is generally practiced outdoors and by hand to the detriment of worker's health [8]. Such a management system tends to be very common in Africa and is integrated into e-waste treatment schemes, combining manual operations and mechanical sorting, among others [9,10].

The presence in this waste of toxic elements such as heavy metals and the damage they represent for the environment and health is not always taken into account by the actors in the sector. Some inappropriate practices of separating electronic equipment are commonly used to extract the precious metals they contain. It is not surprising to see the burning of wire in copper recovery devices, or the use of acidic chemical strippers for the recovery of gold and other precious metals [11,12]. These different practices are likely to release heavy metals in the vicinity of handling sites [13,14]. Trace metals found in electrical and electronic equipment include arsenic (As), cadmium (Cd), chromium (Cr), copper (Cu), mercury (Hg), manganese (Mn), iron (Fe) and lead (Pb), some of which are considered priority pollutants to be monitored regularly due to their persistence and high toxicity [15-18].

Previous studies have shown that these metals can be found in varying concentrations in printed circuit boards of mobile phones and in liquid crystal displays [19-21]. Some metals such as Cu and Fe are known to be essential for human growth and development at low doses [15]. At moderate doses, for example, copper is involved in cellular respiration [22]. However, at high doses, it can cause some liver and kidney damage, respiratory tract irritation and metal fever with prolonged exposure [22,23]. In contrast, heavy metals such as Pb and Cd are known more for their harmful effect than for their health benefits. For example, chronic exposure to cadmium has been implicated in kidney damage and low bone mineral density at the bone level [24]. And the element like Pb, be recognized by the International Cancer Research Agency (IARC) as a probable human carcinogen.

On the other hand, in the environment, the presence of heavy metals in soil can be toxic to plants, animals and humans [25-27]. During prolonged exposure, these trace metals are able to biomagnifying in the environment [28]. Among these metals, Pb, Cd, Cu and Cr are able to negatively influence the yield of agricultural land that has been polluted by them in the vicinity of e-waste recycling sites [11]. However, it is not the total form of the metal in the soil that can be absorbed by the body, if the bioavailable part, i.e. the amount of dissolved fraction or the amount of heavy metal that can physiologically induce bioaccumulation or uptake by plants or other organisms, is absorbed [29].

Despite a growing number of studies already carried out in developed countries on the pollutants contained in e-waste and on the health and environmental impact of these pollutants, these continue to be exported massively to developing countries [30]. In order to highlight the link between the activities of treatment of this electronic waste in the informal environment in Senegal and the contamination of the soil by trace metal elements, this study aimed to determine the levels of metals (As, Cd, Cr, Cu, Fe, Hg, Mn, Pb) in the soil of electronic waste treatment sites in order to : (i) assess the level of contamination of each site by these trace elements, (ii) assess the ecological risk that may result from each of the activities, (iii) estimate the non-carcinogenic and carcinogenic nature of these trace elements at the level of the sites.

Materials and Methods

Study Area and Sampling

This study was carried out in the Dakar region of Senegal. Senegal is a coastal country located in West Africa between 12°80 and 16°41 north latitude and 11°21 and 17°32 west longitude [31]. Three categories of activity sites in the informal environment have been identified: collection sites, recycling sites and dismantling sites (Figure 1).





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Figure 2 shows the location of the sites where these activities were identified and which were used for soil sampling. The soils were collected at 17 sites in eight districts of the Dakar region, namely Rebeuss, Salle de Vente, Medina, Fass, Mermoz, Pikine, Guédiawaye and Thiaroye. The choice of sites was guided by the practice of one of the activities on each of the sites and sampling was done by the snowball method [13]. Collection sites were points where e-waste was collected and stored after being collected on city streets. Recycling sites, on the other hand, were sites where electronic waste was sorted and materials that could be reused were recovered and then transferred to other appliances or sold to the highest bidder. The dismantling sites were those where electronic waste is broken down or broken down for its components (Figures 1 & 2).

A control soil was also collected during the campaign; This witness soil came from the district of Bambilor which is also located in the Dakar region and which had no known history of handling or dismantling electronic waste.

The soil sampling was carried out during the month of June 2021; this month corresponds to the hot season in the Dakar region. The samples were taken at a depth of 0 to 5 cm and once collected, they were stored in plastic bags before being transported to the Laboratory of Toxicology and Hydrology of Faculty of Medicine, Pharmacy and Stomatology of Cheikh Anta Diop University of Dakar for analysis. For each site, 5 soil subsamples were collected and used to obtain a soil/ash/dust composite representative of the site.

Treatment and Analysis of Samples

Once in the laboratory, the samples were dried and then sieved to remove metal debris, pieces of stone and rock, etc. A homogeneous composite representative of each category of activities was then ground and passed through a sieve of different diameters. Fractions smaller than 100 microns of meshes were used for characterization and quantification in metallic trace elements.

Soil Characterization

The methods described by Maiti SK [32] were used for the determination of pH and conductivity. The soil pH was measured after 2 hours of agitation of the soil/doubledistilled water mixture saturated with $CaCl_2$ in a ratio of 1:2.5 (w/v). For conductivity, it was measured in the supernatant of the soil/water mixture e at a ratio of 1:5, after 1 hour of stirring (w/v) [32].

Total Determination of Metals

The total fraction of trace metals (As, Cr, Cd, Cu, Fe, Hg, Mn and Pb) present in different soils was determined using the slightly modified method of the American Public Health Association [33]. Briefly, 0.5 g of soil was treated with 5 mL of nitric acid for trace metal analysis. The resulting mixture was filtered and the final volume was completed at 50 mL and stored at 4 °C until analysis by flame atomic absorption spectrometer.

Bioavailable Fractions of Metals

The bioavailable fraction of trace metals in soil from e-waste sites was determined using the method described by Diop C, et al. [31]. Briefly, 1 g of soil were attacked for 24 hours, at room temperature and with continuous stirring, with 20 mL of 1M HCl. The extraction of the bioavailable fraction of metals by 1 M HCl is comparable to the sum of the first three reactive fractions [31].

Environmental Assessment

The environmental impact of heavy metals was assessed by determining the geoaccumulation index (Igeo), enrichment factor (EF), contamination factor (CF), degree of contamination (Cdeg) and potential ecological risk (RI) of the soil for each study site. For a better understanding of our results in relation to contexts, the geochemical background values that were used are those of the control soil.

Geoaccumulation Index

The Geoaccumulation Index (Igeo) assesses soil contamination by comparing site concentrations with reference or control soil [34]. Originally used for bottom sediments, it can also be applied to the assessment of soil contamination and is determined by the following mathematical equation [35].

$$Igeo = \log_2 \frac{C_n}{1,5B_n} \tag{1}$$

Cn (mg.kg⁻¹) represents the concentration of element n in the soil of e-waste. Bn is the geochemical background (mg.kg⁻¹) of the element n defined by Nazzal Y, et al. [36]. The constant 1.5 represents the influence of deposits and geological features, reduces the effect of human activities, and is used as a correction factor for anthropogenic influence in the calculation. Igeo can be divided into seven classes according to Torabi Kachoosangi's classification [35] (Table 1).

Level	Soil pollution	If	Category
I _{geo} <0	Unpolluted	EF <1.5	No enrichment
$0 \le I_{geo} \le 1$	unpolluted has moderately polluted	1.5 ≤ EF <2	Slight enrichment
$1 < I_{geo} \le 2$	moderately polluted	$2 \le EF < 5$	Moderate enrichment
$2 < I_{geo} \le 3$	moderately to severely polluted	5 ≤ FE <20	Severe enrichment
$3 < I_{geo} \le 4$	severely polluted	20 ≤ <40	Very severe enrichment
$4 < I_{geo} \le 5$	severely to severely polluted	40 <fye< th=""><th>Extremely severe enrichment</th></fye<>	Extremely severe enrichment
5 < I _{geo}	extremely polluted		

Table 1: Classification of accumulation index G (Igeo) and enrichment factor (EF).

Enrichment Factor

Enrichment factor (EF) is defined for soil as the ratio between the relative concentration of an element in a sample and the relative concentration of the same element in a reference environment [34]. Iron can be applied as a reference element in this study, as Senegal's soil is a clayey ferrous soil [37]. Hasan AB, et al. [38] classification (Table 1) were used to locate the level of enrichment of different soils and the following mathematical formula (2) by Torabi Kachoosangi F, et al. [35] made it possible to carry out the calculation:

$$EF = \frac{C_n / C_{fe}}{B_n / B_{fe}}$$
(2)

Cn (mg.kg⁻¹) represents the concentration of element n in the soil studied, Bn (mg.kg⁻¹) represents the concentration of element n in the reference soil. Cfe (mg.kg⁻¹) and Bfe (mg. kg⁻¹) represent the concentration of iron in the sample and the reference soil, respectively [38,35].

Contamination Factor and degree of Contamination

The contamination factor (CF) was determined to assess the overall level of heavy metal pollution in soil samples from the study area. It was determined using equation (3) and classified according to Ngai KW, et al. 2021 [39] (Table 2).

$$CF_n = \frac{C_n}{C_m} \tag{3}$$

Cn (mg.kg¹) represents the concentration of heavy metals in the soil sample and Cm (mg.kg¹) represents the value in the reference soil [39].

The degree of contamination Cdeg was calculated using equation (4).

$$C_{\rm deg} = \sum_{1}^{n} CF_i \tag{4}$$

Cdeg is a parameter for determining the level of pollution of sites by adding the contamination factors ($[CF]_i$) of trace metals found. Contamination factors and degree of contamination were also classified by Shirani et al. 2020 (Table 2).

	Contamination factor	Degree of contamination					
Level	Classification	Level	Classification				
C _i < 1	Low contamination	$C_{deg} < 8$	Low degree				
$1 \le C_{f <} 3$	Moderate contamination	$8 \le C_{deg} < 16$	Moderate degree				
$3 \le C_{f < 6}$	Considerable contamination	16 ≤ C _{deg} <32	Considerable degree				
$C_{f} \ge 6$	Very high contamination	C _{U≥} 32	Very high degree				

Table 2: Classification of contamination factors and degree of contamination (Hakanson 1980).

Potential Ecological Risk Index

The ecological risk assessment of e-waste was conducted by determining the Potential Ecological Risk Index (RI) and is given by the following equation:

$$RI = \sum_{0}^{n} E_{r}^{n} = \sum_{0}^{n} T_{r}^{n} CF_{n}$$
(5)

Where Er is the coefficient of potential risk to the environment and Tr is the toxic response factor (Tr). The Tr

for Cu, Fe, Cd, Mn, Pb, As, and Hg is 5, 1, 30, 1, 5 10 and 40 mg/kg respectively [39,40]. RI and Er were classified according

to the Shirani, et al. [39] (Table 3).

	Не		Re				
Level	Classification	Level	Classification				
$E_r^n < 40$	Low risk	RI <150	Ecological risk				
$40 \le < 80 E_r^n$	Moderate risk	150 ≤ RI 300 <	Moderate ecological risk				
$80 \le < 160 E_r^n$	Considerable risk	300 ≤ RI <600	Significant ecological risk				
$160 \le < 320 E_r^n$	High risk	RI≥ 600	Very high ecological risk				
$E_r^n \ge 320$	Very high risk	-	-				

Health Risk Assessment

Health risk assessments from ingestion, inhalation and dermal exposure were conducted to estimate the potential health consequences of heavy metals [41,42]. Equations (6), (7) and (8) were used to determine the chronic daily intake (CDI) and also to estimate health risks (U.S.EPA 1989; U.S.EPA 2001) [43,42]. The values of the constants are given in (Table 4).

$$CDI_{ing} = C_{bsoil} \frac{IngRxEFxED}{BWxAT} 10^{-6}$$
(6)

$$CDI_{ing} = C_{bsoil} \frac{InhRxEFxED}{PFExBWxAT} 10^{-6}$$
(7)

$$CDI_{ofm} = C_{bsoil} \frac{SAFxSAxABSxEFxED}{BWxAT} 10^{-6}$$
(8)

The health risk in our study was assessed using the hazard index (HI) for non-carcinogenic elements and adding

the individual hazard quotient. The hazard quotient (HQ) was calculated for each route of exposure by dividing the CDI by the reference values (Rfd) of the metals studied. HQ and HI are calculated by the following formulas:

$$HQ_i = \frac{CDI_i}{Rfd} \tag{9}$$

$$HI = \sum_{i}^{n} HQ_{i} \tag{10}$$

The non-carcinogenic risk is accepted when HI < 1. For HI > 1, non-cancer effects could occur, with a probability that tends to increase proportionally to the HI value [44,42]. For carcinogens, the corresponding CDI value is multiplied by the corresponding carcinogenic slope factor (SF) to produce a cancer risk estimate (CR) [45,46]. Total carcinogenic risk (TCR) was obtained by summing the carcinogenic risks for each site. CR and TCR values between 1x10-6 and 1x10-4 indicate acceptable carcinogenic risk, while values above 1x10-4 indicate significant health hazards [45].

Parameters	Symbol	Units	Values	References		
Chronic Daily Intake	CDI	(mg/kg)/day				
Concentration of heavy metals in soil	С	mg/kg				
Average time non- cancer	TA	Days	30 × 365	(U.S.EPA 2001)		
Average time cancer	ATC	Days	70 × 365	(U.S.EPA 2001)		

			Ingestion RfD: 0.0035 (Pb), 0.001 (Cd), 0.04 (Cu), 0.046 (Mn), 0.003 (Cr), 0.0003 (As)	(Kamunda et al. 2016; Liang et al. 2017; Obiri-Nyarko et al. 2021)
Chronic reference dose	Rfd	Per (mg/kg)/day	RfD inhalation: 0.00352 (Pb), 0.000057 (Cd), 0.0402 (Cu), 0.000143 (Mn), 0.0000286 (Cr), 0.000301 (As)	(Liang et al. 2017; Mowla et al. 2021; Obiri-Nyarko et al. 2021; Shashi et al. 2021)
			Dermal RfD: 0.000123 (Pb), 0.0005 (Cd), 0.012 (Cu), 0.00184 (Mn), 0.00006 (Cr), 0.000123 (As)	(Kamunda et al. 2016; Liang et al. 2017; Mowla et al. 2021; Obiri-Nyarko et al. 2021; Shashi et al. 2021)
Carcinogenicity slope factor	Carcinogenicity slope factor SF		SF ingestion: 0.0085 (Pb), 0.5 (Cr), 1.5 (As)	(Kamunda et al. 2016; Liang et al. 2017; Obiri-Nyarko et al. 2021)
			SF inhalation: 0.042 (Pb), 6.3 (Cd), 290 (Cr), 15.1 (As)	(Affairs 2010; Kamunda et al. 2016; Liang et al. 2017; Obiri- Nyarko et al. 2021)
			SF Ingestion:	(Kamunda et al. 2016; Liang
			0.0000085 (Pb), 3.66 (As)	et al. 2017; Obiri-Nyarko et al. 2021)
Body weight	BW	Kg	67	
Dermal absorption factor	ABS	Unitless	0.1	(Affairs 2010)
Exposed skin exposed surface area	HIS	m ²	5800	(Affairs 2010; Kamunda et al. 2016)
Exposure duration	ED	Year	30	(U.S.EPA 2001)
Exposure frequency	EF	Day/year	350	(U.S.EPA 2001)
Skin adherence factor	SAF	mg/cm ²	0.07	(U.S.EPA 2001)
Ingestion spleen	IngR	mg d ⁻¹	100	(U.S.EPA 1989; U.S.EPA 2001)
Inhalation spleen	InhR	m ³ /day	20	(U.S.EPA 2001)
Particular emission factor	PEF	m ³ /kg	1.36E-09	(U.S.EPA 2001)

Table 4: Parameters and values of constants used for risk assessment.

Statistical Analysis

Data were recorded and processed statistically using Microsoft Excel 2019 and SPSS 24.0.12 (Statistical Package for the Social Sciences). Statistical analyses of handling sites were presented as mean deviations \pm type. Pearson's analytical correlation method was used for the multi-pair comparison of the level of correlation between heavy metals.

Results and Discussion

Sampling

During the campaign, seventeen sites dealing with e-waste were identified in eight neighbourhoods in the Dakar

region. Five of these sites specialized in e-waste collection, five in recycling and the last seven in dismantling (Figure 2).

Given the climate and the many movements that take place every day on the sites, the diffusion and leaching of trace metals during the dismantling and landfilling processes of WEEE recycling can affect several parameters, including hydrogen potential and soil electrical conductivity [47].

Potential of Hydrogen and Conductivity of Soil

The physicochemical properties of the soils analysed are presented in Table 5. Our results place the pH of soils between 6.21 and 7.6 1, which classifies them as low acid to low alkaline. Among e-waste handling sites, collection sites had the highest pH (7.08), followed by recycling (6.81) and dismantling (6.69). As in our study, low acidic pH values were also found by Amphalop N, et al. [11] in Thailand. However, the pH values found here for dismantling and recycling sites are slightly lower than that obtained by Isimekhai KA, et al. [48] in Nigeria. Indeed, in his study on the distribution of heavy metals in recycling sites in the informal sector, Isimekhai KA, et al. [48] found a pH between 7.08 and 7.52 for recycling sites, and between 7.71 and 8.32 for dismantling sites [48]. pH is one of the most important factors affecting the diffusion and potential ecological risk of heavy metals [49]. In general, the mobility of metals in the environment increases with decreasing soil pH due to precipitation of metals in the form of hydroxides, carbonates or the formation of insoluble organic complexes [49-51]. Unlike the soils studied by Isimekhai KA, et al. [48] in Nigeria, the low acid nature of the dismantling and recycling sites studied here suggests that some metals such as Pb, Cd will see their toxicity increase in these two types of soils compared to collection sites.

The mean values of the electrical conductivity are generally similar for all handling sites, and higher than that of the control site (1.82 mS.cm⁻¹) (Table 1). This indicates the presence of a large amount of inorganic substances dissolved in ionized form [52]. These conductivity results in e-waste handling sites were higher than those found in solid waste

landfills in Sri Lanka which were around 1136 $\mu S.cm\text{-}1$ and 3720 $\mu S.cm\text{-}1$ respectively in the water adjacent to the landfill [53].

Concentration and Distribution of Heavy Metals in Site Soils

The determination of trace element concentrations in samples from different sites has a wide range of concentrations that varied depending on activity (Table 5). In general, the abundance of metals selected for collection and recycling sites was in this order Fe> Mn> Pb> Cu> Cr> Cd> As> Hg. On the other hand, at the dismantling sites, the total concentration of these elements was classified according to whether Fe> Pb> Mn> Cu> Cr> As> Cd> Hg and that of the control site were Fe> Mn> Cu> Pb> Cr> Cd> As> Hg. With the exception of Fe and Mn whose total average concentrations (Fe: 7599.03 mg.kg-1; Mn: 802.15 mg.kg⁻¹) the highest was obtained in the recycling sites, the dismantling sites had the highest total concentrations for all other metals and they were between 679.85 and 830.83 mg.kg⁻¹ for Pb, 265.38 and 354.51 mg.kg⁻¹ for Cu, 48.24 and 19.10 mg.kg⁻¹ for Cr and between 4.29 and 14.09 mg.kg⁻¹ for the Cd (Table 2). Hg was the trace element with the lowest concentration at all sites.

Devenuetova	Statistics		Sai	nple sites		
Parameters	Statistics	Control	Collector	Dismantling	Recycling	
h	Mean	7.01 ±0.67	7.08 ±0.32	6.69 ±1.33	6.81 ±0.38	
рп	Range		6.77 – 7.61	6.54 - 7.01	6.21 ±7.25	
$\mathbf{E} \in (\mathbf{mS} \times \mathbf{m})$	Mean	1.82	20.31 ±2.61	20.20 ±1.33	20.71 ±1.50	
E.C. (Range		17.69 - 23.12	19.06 - 22.80	19.32 - 23.01	
	Mean	4	237.00 ±101.12	782.50 ±36	476.27 ±136.22	
Pb (mg/kg)	Range		35.45 - 387.31	679.85 - 830.83	336.45 ±724.08	
	% bioa	44.98	61.17	47.85	49.76	
	Mean	1.82	3.94 ±1.06	7.39 ±3.57	6.18 ±1.34	
Cd (mg/kg)	Range		2.57 - 5.51	4.29 - 14.09	4.39 - 8.72	
Cd (mg/kg)	% bioa	15.34	58.47	77.82	49.76	
	Mean	27.97	217.22 ±94.75	337.92 ±32.18	324.66 ±14.59	
Cu (mg/kg)	Range		63.83 - 288.41	265.38 - 354.51	288.18 - 345.29	
	% bioa	58.54	54.64	84.61	71.9	
	Mean	28.97	418.81 ±183.39	731.70 ±310.15	802.15 ±275.60	
Mn (mg/kg)	Range		126.88 - 604.38	283.29 -1238.01	383.41-1371.70	
	% bioa	74.46	26.8	37.89	32.37	

	Mean	754.59	6800.85 ±1105.31	7475.77 ±294.88	7599.03 ±191.54	
Fe (mg/kg)	Range		4937.81 - 7753.45	6968.47-7957.13	7163.58-7966.62	
	% bioa	32.18	65.84	74.45	72.95	
	Mean	2.37	34.64 ±16.33	38.43 ±11.43	40.90 ±5.18	
Cr (mg/kg)	Range		13.44 - 57.98	48.24 - 19.10	31.64 - 51.35	
	% bioa	24.41	43.45	41.04	38.09	
As (mg/kg)	Mean	0,49	1.50 ±1.28	10.13 ±8.77	0.63 ± 0.41	
	Range		0,25-3,49	0,12-22,42	0,11-1,25	
	% bioa	12.37	31.78	40.61	22.05	
Hg (mg/kg)	Mean	0,029	0.29 ±0.24	0.84 ±0.30	0.38 ±0.19	
	Range		0.16 - 0.73	0,45-1,31	0,18-0,65	
	% bioa	12.81	29.5	67.05	21.46	

Table 5: Physicochemical properties and mean concentrations of trace elements in soil (mg/kg). % bioa: Percentage of bioavailable fraction

The concentrations of trace elements found at soil levels may be influenced both by the particle dispersion pattern and by soil heterogeneity. For our sites, the concentrations obtained are all higher than that of the control site. These results corroborate those obtained in Nigeria and China in the soils of e-waste handling sites, suggesting that the long-term handling of e-waste is one of the main causes of the increase in soil pollution by heavy metals such as cadmium, lead and copper [54-56]. In the Alaba e-waste landfill in Nigeria, the Pb value (1535 mg.kg⁻¹) was 2 times higher than the average concentration of dismantling sites, 3 times higher than the average concentration of recycling sites and 6 times to those of Dakar collection sites [54]. The Cd at Alaba, on the other hand, had a value (7.69 mg/kg) which was closest to the average value of the dismantling sites $(7.39 \pm 3.57 \text{ mg/kg})$ and that of the recycling sites (6.18 mg/kg) in Dakar. These various observations highlight the influence of activities on the concentration of some trace elements such as Pb. In addition, these results are explained by the fact that some metals are widely used in the manufacture of electronic products and equipment that are handled at the site level. Pb and Cd, for example, are present in printed circuit boards, Cd in computer batteries, and Cu in electrical cables, and these are readily released into the environment during the handling of these devices [57-59]. For this reason, some authors consider that landfills and e-waste handling sites play a significant role in the anthropogenic input of traces metals elements into soil [14,26,60-62].

The percentages of bioavailable fractions are also shown in (Table 2). In the different sites, Cd, Cu, Pb, Mn, Cr and Fe have a bioavailability percentage between 26.80 and 84.61%, suggesting high mobility and bioavailability of these elements. Of the metals studied, Mn and Cr were the least bioavailable elements for each site category. With the exception of Pb, dismantling sites had the highest percentage of bioavailability for other elements. This can be explained by the pH and conductivity values of these sites. In addition, an increase in salinity is associated with an increase in concentrations of elements such as Na, K, Ca, Mg mostly competing with heavy metals, which may contribute to a greater release of exchangeable metals and an increase in the mobility and availability of trace metals [63].

In order to better assess the impact of these polluting activities on the soil and the resulting ecological risk, several parameters have been identified and evaluated.

Geoaccumulation Index (Igeo), Enrichment Factor (EF), Degree of Contamination (Cdeg) and Potential Ecological Risk Index (RI) of Heavy Metals in Site Soils

Geoaccumulation Index and Enrichment Factor: The geoaccumulation index (Igeo) and the enrichment factor (EF) made it possible to assess the degree of metal pollution of the soils of the study sites. Igeo is an effective indicator for understanding and assessing the level of trace metal pollution [64]. This is usually done by comparing the background value to determined metal concentrations, or from a reference soil [34,65]. In this study, trace element concentrations for the control site were used to assess the impact that the handling of e-waste can have on soil and health. This choice makes it possible to compare the pollution on the exposed sites with that of a site that is not.

According to the classification of Torabi Kachoosangi F, et al. [35] the data presented in Table 3 allow to apprehend the overall contamination average by category of site. Of these,

the dismantling sites were the most contaminated by the elements Pb, Cd, Mn, As and Hg. More specifically, these sites were « extremely polluted » by Pb, « severely to extremely polluted » by Hg, « moderately to severely polluted » by Cd and As, and the average values of Igeo were 7.03, 4.21, 2.25 and 2.57, respectively (Table 6). The average Mn pollution was classified as « severely polluted ». As for Cu, Fe and Cr, the highest average level of pollution by these elements was

obtained in recycling sites and the Igeo values were 3.04 for Cu, 2.73 for Fe and 3.41 for Cr, which classified the average soil pollution by Cu and Cr as « severely polluted » and Cu to « moderately to severely polluted ». The lowest overall level of contamination was that of As and it was obtained in the collection and recycling sites and the average value of the Igeo of the As in these sites was less than 1.

Samples				Ig	eo					EF						
		Pb	CD	Cu	Mn	Fe	Cr	As	Hg	Pb	CD	Cu	Mn	Cr	As	Hg
Collector	Mean	4.94 ±1.39	1.43 ±0.40	2.19 ±0.92	3.09 ±0.90	2.57 ±0.026	3.13 ±0.78	0.49 ±1.45	2.50 ±0.90	7.11 ±4.71	0.46 ±0.11	0.83 ±0.30	1.54 ±0.55	1.58 ±0.67	0.34 ±0.28	1.32 ±1.44
	Range	2.56- 6.01	0.85- 1.95	0.61- 2.78	1.55- 3.80	2.12- 2.78	1.92- 4.03	-1,58– 2,22	1,91- 4,08	0.86 - 12.60	0.38 - 0.65	0.35 – 1.08	0.67 – 2.03	0.87 – 2.61	0,05– 0,78	0,57– 3,89
Dismantling	Mean	7.03 ±0.10	2.25 ±0.64	3.00 ±0.15	3.95 ±0.69	2.72 ±0.06	3.36 ±0.51	2.57 ±2.74	4.21 ±0.55	19.80 ±1.59	0.78 ±0.36	1.22 ±0.1 2	2.53 ±1.01	1.63 ±0.47	2.07 ±1.77	2.95 ±0.98
	Range	2.56- 7,12	0.85- 3.31	0.61- 3,08	1.55- 4,83	2.13- 2,81	1.92- 3,76	-2,58– 4,90	3,39– 4,94	17.03 - 21.34	0.45 - 1.48	0.97 – 1.36	0.98 -4.05	0.81 – 2.09	0,02- 4,51	1,71- 4,36
Recycling	Mean	6.92 ±0.22	2.20 ±0.73	3.04 ±0.03	3.89 ±0.71	2.73 ±0.01	3.41 ±0.58	-0.62 ± 1.26	2.99 ±0.78	11.77 ±3.76	0.65 ±0.17	1.15 ±0.04	2.72 ±1.17	1.72 ±0.33	0.12 ±0.07	1.32 ±0.66
	Range	6.57- 7,09	1.59- 3,30	2.99- 3,08	2.70- 4,46	2.71- 2,74	2.42- 3,85	-2,66– 0,73	2,03– 3,93	8.24 – 17.17	0.46 – 0.92	1.09 – 1.18	1.39 – 4.49	1.26 – 2.15	0,02- 0,24	0,65– 2,28

Table 6: Igeo and enrichment factor (EF) for the different site groups.

The Igeo values of these sites show soil contamination of anthropogenic origin. Informal recycling and horizontal spread of metal contaminants are factors that can contribute to the diffusion of metals in handling and dismantling areas [30,34]. Soil pollution observed at dismantling sites can be promoted by the combustion of equipment wires and cables [11,55,56]. Our results also corroborate those of Ackap 2019, which located the arable land and subsoil of the Agbogblohie and Ashaiman e-waste dismantling sites as « highly contaminated» to « extremely contaminated » with Pb, Cu and Cd [66].

In addition to Igeo, the analysis of enrichment factor (EF) values shows an EF that ranges from deficiency to severe soil enrichment, with an average EF of lead that was in the category of « severe enrichment » for all sites (Table 3). This observation again indicates an influence of anthropogenic e-waste handling activities as a source of Pb compared to the control site. Similar soil enrichment by Pb has already been observed at Agbogbloshie in Ghana, where the level of Pb enrichment was classified as « severe » at the waste dismantling sites [67]. In addition, elements such as Pb and Cr may have been introduced into the soil via certain equipment such as batteries, fluorescent lamps, cable sheaths and plumbing materials [51]. In addition, at the dismantling sites, there is a significant « moderate enrichment » of the

soil in As and Hg (Table 6). Since the determination was made using data from a site where no industrial activity is carried out, the different levels of enrichment that have been achieved appear to be due to the treatment of e-waste on these soils [34,62,68].

These results from Igeo and EF that have been obtained make it possible to understand the influence of anthropogenic activity of handling these wastes on the soil [2,31].

Contamination Factor and Degree of Soil Contamination

Based on the results obtained from the contamination factor and degree presented in (Table 7), with the exception of Cd and As in collection sites and As in recycling sites which had a contamination factor below 6, all sites had "Very High Contamination" and therefore « Very High Degree » of average contamination at soil level with respect to other elements compared to the control soil. These results indicate a significant contribution of anthropogenic sources to soil pollution at our sites [51]. Among these pollutants, Pb is responsible for a high degree of soil contamination with an average input of 48.39% at collection sites, 61.82% for dismantling sites and 51.67% for those of recycling.

				C	f									Er					DI
Samples		Pb	Cd	Cu	Mn	Fe	Cr	As	Hg	Ldeg	Pb	CD	Cu	Mn	Fe	Cr	As	Hg	RI
	Auguara	59.33	4.15	7.77	14.46	9.01	14.60	3.00	10.28	122.60	296.67	124.54	38.84	14.46	9.01	29.21	29.98	411.23	953.93
	Average	±34.16	±1.12	±3.39	±6.35	±1.46	±6.89	±2.57	±8.50	±32.62	±170.79	±33.49	±16.94	±6.35	±1.46	±13.77	±25.69	±340.19	±390.90
collector	Range	8.87 -	2.71 -	2.28 -	4.38 -	6.54 -	5.66 -	0,50-	5,65-	76,54-	44.37 -	81.34 -	11.41 -	4.38 -	6.54	11.33	5,00-	225,96-	557.21
	Kalige	96.94	5.80	10.31	20.87	10.27	24.44	6,99	25,46	160,83	484.72	174.10	51.56	20.87	10.27	48.88	69,85	1018,25	1560.66
	Freil	195.86	7.79	12.08	25.26	9.91	16.20	20.28	29.45	316.84	979.30	233.72	60.41	25.26	9.91	32.40	202.82	1178.20	2722.02
Dia	EVII	±12.77	±3.76	±1.15	±10.71	±0.39	±4.82	±17.55	±10.60	±33.09	±63.85	±112.93	±5.75	±10.71	±0.39	±9.64	±175.54	±423.94	±482.51
mantling	Range	170.17 -	4.52 -	9.49 -	9.78 -	9.23 -	8.05 -	0.25 -	15.75 -	264.75 -	850.83 -	135.61	47.45 -	9.78 -	9.23 - 1	16.11 _	2.50 -	630.18 -	2090.86 -
	runge	207.96	14.85	12.68	4 2.74	10.54	20.34	44.90	45.98	360.47	1039.79	445.36	63.38	42.74	0.54	40.68	448.97	1839.30	3288.50
	Enil	119.21	6.51	11.61	27.69	10.07	17.24	1.25	13.35	206.93	596.06	195.39	58.04	27.69	10.07	34.49	12.47	533.89	1468.10
	EVII	±41.55	±1.79	±0.77	±12.81	±0.39	±3.09	±0.83	±6.80	±57.22	±207.74	±53.68	±3.85	±12.81	±0.39	±6.18	±8.31	±272.19	±444.08
Recycling	Range	84.21 -	4.63 -	10.30	13.24 -	9.49 -	13.34 _	0.24 -	6.12 -	153.05 -	421.06 -	138.93 -	51. 52-	13.24 _	9.49 -	26.68 -	2.38 -	244.91 -	980.43 -
		181.24	9.19	12.35	47.36	10.56	21.65	2.49	22.91	291.35	906.19	275.76	61.73	47.36	10.56	43.30	24.94	916.49	1964.67

Table 7: Summary of contamination factors, contamination indicator, maximum ecological risk coefficient and ecological risk for different site groups.

FC: contamination factors; C_{deg}: to the extent of contamination; Er: lateral ecological risk coefficient; PERI: potential ecological risk index.

Potential Ecological Risk Index

The potential ecological risk coefficient (Er) and the potential ecological risk index (RI) were determined to individually understand and aggregate the potential ecological risk of the metals Pb, Cd, Cu, Fe, Mn, Cr, As and Hg in soil relative to the control site, and the results are reported in the table 5. The average Er values for Fe, Mn and Cr were less than 40 for all sites, as well as As in collection and recycling sites and Cu in collection sites. These results therefore indicated a low potential ecological risk to aquatic organisms from these metals [28]. Pb, Cd and Hg were average level of « Very High Risk » ecologically for all sites. The As also had a « Very High Risk » for dismantling sites. Thus, from these results, the resulting ecological risk index places each of these sites at a « Very High Ecological Risk » (RI> 600) index. Therefore, the e-waste handling sites that have been assessed pose a threat to the surrounding ecosystem and a high ecological risk due to the activities carried out [25]. Similar results were obtained at the e-waste landfills in Qingyuan in China, Guwahati in India and the Kpone landfill in Ghana [25,51,69]. In these studies, and as in our study, Hg, Pb and Cd contribution to the level of ecological risk. This can be explained by the role played by leaching or by acid combustion at these sites, which play a substantial role in pollution, unlike abandoned sites and low-activity sites with

much lower values [12]. Moreover, the isolated extraction of the metals of interest would not have provided a real insight into the RI. Indeed, the Pb represents respectively 31.09%, 35.98%, 40.60%, of the RI of the sites, collection, dismantling and recycling. While Hg represents 43.11%, 43.28% and 36.37%. With a contribution of more than 70% from these two elements, this makes it possible to classify them as elements that require special monitoring with regard to ecological and toxicological risks to the environment at these sites. Hence it would be wise to say that these two metals contribute significantly to the increase in ecological risk compared to other metals [49]. It should also be noted that ecological and health risk assessment is generally quite complex and can be influenced by many factors, such as topography, presence of organic matter, hydrological conditions and interactions between the different elements present at each site [70].

The results of Pearson's correlation analysis Table 8 show a significant positive correlation of 0.05 levels between lead and cadmium, copper and arsenic, which assumes that these metals would come from the same source. However, iron showed a strong positive correlation at 0.01 with copper and manganese, which may indicate a different source than other heavy metals [27,44]. In addition, the strong correlation between some metals indicates the presence of generally anthropogenic sources of contamination [44].

Variable	Pb	Cd	Cu	Mn	Fe	Cr	As	Hg
Pb	1							
Cd	0.502*	1						
Cu	0.492*	0.484*	1					
Mn	0.477	0.459	0.572*	1				
Fe	0.233	0.407	0.862**	0.656**	1			
Cr	0.112	0.319	0.37	0.44	0.478	1		
As	0.526*	0.121	0.363	-0.048	0.07	-0.12	1	
Hg	0.372	0.389	0.274	0.500*	0.204	0.13	0.276	1

Table 8: Pearson correlation between heavy metal concentrations and physicochemical properties.

* The correlation is significant at the level of 0.05 (p < 0.05); ** The correlation is significant at the 0.01 level (p < 0.01).

								×
		HI		но		Cr		Tom
	Ingestion	Inhalation	Dermal	пұ	Ingestion	Inhalation	Dermal	
Control	4.70E-03	1.22E-05	3.83E-02	4.30E-02	1.21E-06	6.34E-08	1.30E-06	2.58E- 06
Collector	7.09E-02	1.63E-04	1.14E+00	1.21E+00	1.33E-05	9.13E-07	3.91E-06	1.81E- 05
Dismantling	2.05E-01	2.28E-04	3.16E+00	3.37E+00	2.52E-05	1.03E-06	2.64E-05	5.27E- 05
Recycling	1.21E-01	2.29E-04	2.06E+00	2.18E+00	1.56E-05	1.08E-06	1.63E-06	1.83E- 05

Table 9: Non-cancer and carcinogenic risk associated with occupational exposure of adults.

Non-cancer and Carcinogenic Health Risk Assessment

Many metals such as Mn, Cu and Fe are considered non-carcinogenic by the International Agency for Research on Cancer [71]. As exposure took place in an occupational environment, the assessment of the carcinogenic and noncarcinogenic risk was made taking into account only the adult population. With regard to carcinogenic risk, only As, Pb, Cd and Cr were assessed.

As shown in Table 9, mean dermal HI values were elevated compared to the digestive and respiratory tracts. With an average HQ greater than 1, dermal exposure to these soils presented relatively greater non-cancer risks, regardless of the type of activity related to the handling of e-waste. This mean dermal HI had a direct consequence of the total hazard quotient for each group. This represents respectively 94.13%, 93.91% and 94.43% of the total HQ value for collection, dismantling and recycling sites. With a total HI value greater than 1, individuals exposed during these activities have a chance of developing non-cancer effects, with the likelihood that they have increased with HI [42,44]. Unlike our results and those of Wei J, et al. [72]

which show a non-carcinogenic risk in the order HIDer > HIIng> HIInh, some studies such as those of Mowla M, et al. [17] and Singh M, et al. [26] had found as the main route of exposure to non-cancer health risk from ingestion exposure, with a significant contribution of Pb [17,26,72]. According for some previous results, the mean values for inhalation exposure show that this route has a significantly higher risk of potential harm to human health [51,26]. For our results, sites where dismantling takes place are at higher risk than others. On the other hand, according to our results, dermal exposure is likely to cause more damage than other exposure routes.

The carcinogenic risk in this study was assessed for Pb, Cd, Cr and As because of their toxicity levels. The mean values of Tcr that were obtained for these elements at the individual sites ranged from 1. 81x10–5 and 5.83x10–5. Although this carcinogenic risk did not exceed the acceptable level (<10-4), this study did not take into account certain metals such as nickel and tin, which could increase the carcinogenic risk [72]. Moreover, from these results, it should be said that the populations living near waste treatment sites present a carcinogenic risk proportional to the environment to which they belong. In this context, children, who are much more

sensitive, would be the most exposed, especially in the event of exposure to lead, which already poses a risk to adults [72-76].

Conclusion

Our study examined the content of metallic trace elements on the soils of e-waste treatment sites in Dakar, Senegal. For this, the study sites were classified into three groups according to the activity. The concentrations of the different elements were all higher at the different sites than at the control site, indicating an anthropogenic origin of the observed contamination. The study indicated that soils at e-waste treatment sites are heavily polluted by Pb and Hg and the accumulation of metals in the soil was found to be proportional to the nature of the activity carried out. In most cases, the dismantling of electronic waste contributes the most to soil accumulation and enrichment. In addition, one of the consequences associated with the dispersion of heavy metals in the soil is the ecological risk they represent. Among the elements that require strict monitoring, Pb and Hg stand out with an important contribution compared to the other elements. Although the carcinogenic risk values are within normal limits, the non-carcinogenic risk is always permanent. The synergistic action of heavy metals in the body can have an impact on the health of the population and especially on children. Thus, these results highlight the need to optimise technologies for recycling and dismantling electronic waste in order to remedy the pollution that would emanate from these activities.

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