

Toxicity and Bioaccumulation Studies of Heavy Metals on a Freshwater Fish

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Abstract

Heavy metals are stable and persistent environmental contaminants of aquatic environments. Toxicological survey was done to investigate the toxicity of selected hazardous heavy metals and also to determine the rates and levels of bioaccumulation attainable in *Clarias gariepinus* when exposed to single, binary and multiple mixtures of heavy metals in laboratory bioassays. There were significant departures when the toxicity levels of mixtures (binary and multiple) were compared to the toxicity levels of the individual metals when acting alone against the same test animal. When most of the mixtures were tested against the test *Clarias gariepinus*, the interaction between the constituents was mainly in conformity with the model of synergism while only a few cases conformed to the model of antagonism. Exposure of the test animal *Clarias gariepinus* to sublethal concentration of the metal mixtures under the joint action studies resulted in a reduction in the concentration of Cd, Pb, and Cr accumulated by the test organism, when compared to the concentrations accumulated by the animal during the single action studies. Therefore, *Clarias gariepinus* was found to bioaccumulate heavy metals (Cd, Cr and Pb) to varying degrees, dependent on the type of metals, period of exposure, and concentration of metal compound in the test media and the joint action of the metals in the system.

Keywords: Heavy Metals; Toxicity, *Clarias gariepinus*; Bioaccumulation; Synergism; Antagonism

Introduction

Heavy metals are natural constituents of the marine and freshwater environment, generally found in very low concentrations. Generally, the expression 'Heavy Metals' is used where there are connotations of toxicity. A number of authors have critically reviewed the usage of

the term "Heavy Metals" and have called it hopelessly imprecise and objectionable [1]. Heavy metals are dangerous to aquatic organisms and it can be bioaccumulated in the food chain leading to diseases in human. They occur in the environment both as a result of natural processes and as pollutants from human activities. According to the literatures, heavy metal bioaccumulation

by fish and subsequent distribution in organs is greatly interspecific. In addition, many factors can influence metal uptake like sex, age, size, reproductive cycle, swimming pattern, feeding behavior, and geographical location [2]. For these reasons, evaluation of heavy metal levels in commercially important fish is important from a toxicological perspective, verifying their effects on the natural environment and significant health risk arising from fish consumption.

Materials and Methods

Test Animals

Fish samples (*Clarias gariepinus*) were collected for toxicological studies from a fish farm and transported in glass tank to the laboratory.

Bioassay Studies

After acclimatization, the fish samples were randomly assigned to bioassay containers. For the series of bioassays, 20 *Clarias gariepinus* were exposed per treatment in two replicates (10 fish per replicate). There was a control in all the treatments. In these bioassays after range finding preliminary trials, fish samples were exposed to series of concentrations of each heavy metal compounds.

A series of bioassays were carried out, but this time, the acclimatized animals were exposed to different concentrations of binary and multiple mixtures of heavy metal salts in pre-defined ratios. There was always 10 *Clarias gariepinus* per bioassay container exposed at each concentration or treatment. Each treatment was replicated 2 times, meaning that 20 *Clarias gariepinus* were exposed per concentration of heavy metal mixtures as specified below.

In all bioassays, mortality data were taken once every 24h over the 96h duration. Test animals were taken as dead if they failed to respond by moving away when touched gently with a glass rod.

Bioaccumulation Studies

Clarias gariepinus was exposed to sublethal concentration of single metal compounds (Pb, Cr, Cd) in semi-static bioassays. In this series of experiments, *Clarias gariepinus* was exposed to only sub-lethal concentrations ($1/10^{\text{th}}$ and $1/100^{\text{th}}$ of 96-h LC_{50}) of the selected metal salts acting singly as specified below. A total of 60 test animals were exposed per sublethal concentration in 3 replicates (20 animals per replicate). In

this series of bioassay that went on for 28 days in order to investigate the rate of bioaccumulation, the semi-static bioassay procedure was always adopted to avoid drastic changes in concentration of test media via evaporation and excessive reduction in dissolved oxygen level. In this semi-static procedure, each test medium was changed into a fresh solution at exactly the same concentration of heavy metal salt once every 7 days, transferring the same exposed test animals into the freshly prepared test media over the 28-day period of the experiment. At predetermined time intervals (day 0, 7, 14, 21 and 28), One live *Clarias gariepinus* per replicate, making three per treatment were randomly selected, cleaned thoroughly with distilled water and placed in labeled polythene bags in which they were kept frozen awaiting digestion of the extracted whole animal tissue and analysis for test metals by AAS [3].

Bioaccumulation of heavy metals by *Clarias gariepinus* was also exposed to sublethal concentrations of binary and multiple heavy metals compounds in mixtures. In order to evaluate the possible influence of ionic competition for uptake sites or other modes of possible interactions on rates and amount of bioaccumulation of metals by animals exposed to mixtures of metallic elements, the following series of bioassays were carried out in the laboratory. A similar semi-static bioassay as described above was carried out, but for this present bioassays, *Clarias gariepinus* was exposed to sublethal concentrations of binary mixtures of Cd - Pb, Cd - Cr, Cr - Pb and multiple mixtures of Cd - Cr - Pb. The binary and multiple mixtures consisted of sublethal concentration of each constituent metal salt, the computation of which was based on pre-determined fractions ($1/10^{\text{th}}$ and $1/100^{\text{th}}$) of the 96-h LC_{50} values of the test metals obtained in acute toxicity of joint action experiments conducted in this work.

Statistical Analysis

Toxicological data based on quantal response (mortality) was analysed by probit analysis, after Finney (1971). The probit analysis, dependent on maximum likelihood interactive regression was done by a computer programme designed and implemented by Ge le Patourel, imperial College London and adopted by Don-Pedro [4]. The above analysis when it involved mortality-dosage data resulted in derivation of LC_{50} , LC_{95} and LC_5 with their 95% confidence limits and other relevant parameters, while when mortality time was analyzed, LC_{50} values were obtained. Pooled cumulative quantal response (mortality) data (of replicates/treatment) was used in probit analysis following accepted toxicological procedures [4].

TF (Toxicity Factor): This was used to measure the relative potency ratio.

RPR = $\frac{LC_{50} \text{ of a compound X}}{LC_{50} \text{ of another compound Y}}$

Where X is the metal with higher value of LC_{50} and Y is the metal with the lowest value of LC_{50} . Histograms were used to compare the LC_{50} values of the metals when acting singly.

Analysis of Data and Measurement of Joint Action Toxicity of Binary and multiple Mixtures of Test Compounds

Joint action toxicity was determined and analyzed using synergistic ratio model, according to Hewlett and Plackett (1969) as follows:

S.R (Synergistic ratio) = $\frac{LC_{50} \text{ of a chemical acting alone}}{LC_{50} \text{ of chemical + additive mixture}}$

Translation of Results

Where S. R = 1 – describes additive action
S.R = < 1 – describes antagonism
S.R = > 1 – describes synergism

Significance /overlap in 95% confidence limit of the detected 96h LC_{50} values were determined using the Chi-Square technique. The limit of significance was 0.05.

Results

Single Acute Toxicity of Cadmium (Cd) Against *Clarias gariepinus*

The 96-h LC_{50} values of Cadmium against *Clarias gariepinus* are shown in 1. The values were calculated from 24h – 96h of the study duration. There was no significant difference observed among the 96-h LC_{50} values of the fish species.

Time	LC_5	LC_{50}	LC_{95}	Toxicity factor (TF)
24h	4.443	11.253	28.505	1.36
48h	3.493	10.146	29.472	1.23
72h	3.692	8.773	20.843	1.06
96h	3.794	8.280	18.071	1.00

Table 1: Acute toxicity of cadmium acting singly against *Clarias gariepinus* (mg/L).

Single Acute Toxicity of Lead (Pb) Against *Clarias gariepinus*

Table 2 presents the 24h – 96-h LC_{50} values of lead

against *Clarias gariepinus*. There were no observed significant variations ($P > 0.05$) on the 24-h LC_{50} , 48-h LC_{50} , 72-h LC_{50} and 96h LC_{50} values.

Time	LC_5	LC_{50}	LC_{95}	Toxicity factor (TF)
24h	38.732	94.545	230.788	1.35
48h	28.745	84.285	247.140	1.20
72h	28.941	73.714	187.757	1.05
96h	29.540	70.183	166.745	1.00

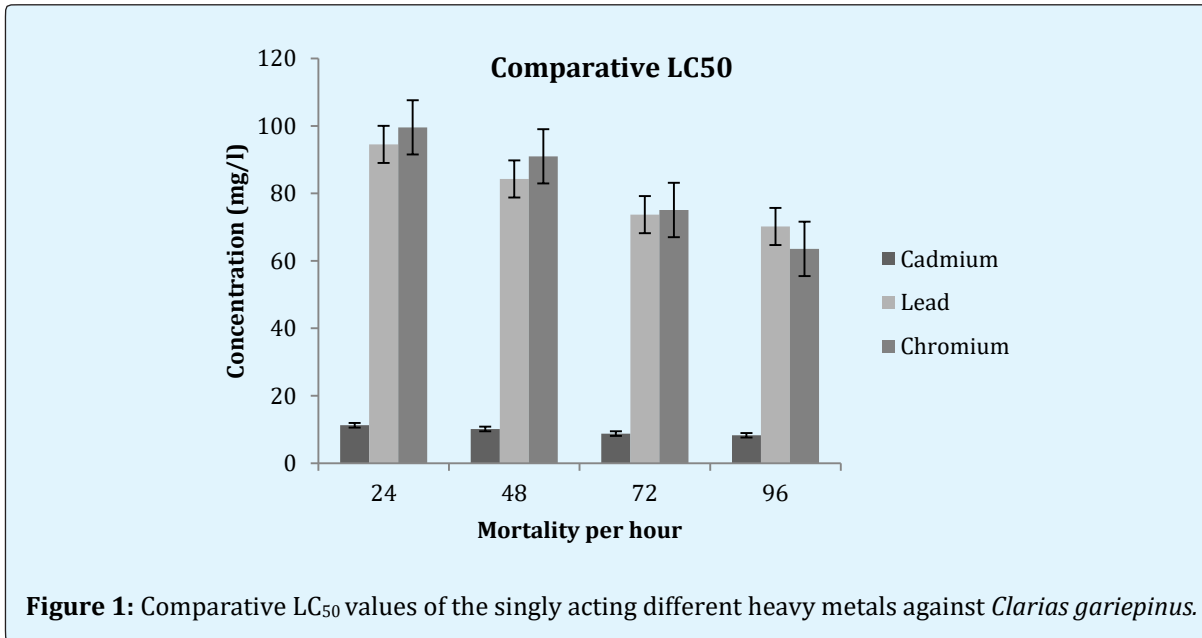
Table 2: Acute toxicity of Lead acting singly against *Clarias gariepinus* (mg/L).

Single Acute Toxicity of Chromium (Cr) Against *Clarias gariepinus*

The 96-h LC_{50} values through 24h duration of fish species exposure are shown on Table 3. The LC_{50} values for the 24-96h exposure recorded no variation ($P > 0.05$).

Time	LC_5	LC_{50}	LC_{95}	Toxicity factor (TF)
24h	36.394	99.577	272.450	1.57
48h	21.257	90.995	389.517	1.43
72h	26.888	75.077	209.631	1.81
96h	19.986	63.546	202.049	1.00

Table 3: Acute toxicity of Chromium acting singly against *Clarias gariepinus* (mg/L).



The numerical comparative chart of the LC₅₀ values of Cr, Pb and Cd studied against the *Clarias gariepinus* is shown on Figure 1. From the figure above, cadmium (Cd) was the most toxic metal and lead (Pb) the least toxic metal to the test animals based on 96-h LC₅₀.

Binary Acute Toxicity of Cadmium (Cd) and Lead (Pb) Against *Clarias gariepinus*

The relative joint acute toxicity (24h - 96-h LC₅₀ values) of the ratios of mixtures for cadmium and lead against the catfish are presented in Table 4.

Time	LC ₅	LC ₅₀	LC ₉₅	Toxicity factor
24h	4.158	38.093	349.025	2.15
48h	4.158	38.093	349.025	2.15
72h	2.608	25.129	242.124	1.42
96h	2.342	17.699	135.753	1.00

Table 4: Acute toxicity of binary mixture of Cadmium (Cd) and Lead (Pb) against *Clarias gariepinus* (mg/L).

Binary Acute Toxicity of Cadmium (Cd) and Chromium (Cr) against *Clarias gariepinus*

The 24-h LC₅₀, 48-h LC₅₀, 72-h LC₅₀ and 96-h LC₅₀ values of binary combination of Cd and Cr in a specified

ratio against *Clarias gariepinus* are shown on Table 5. The 24 -96-h LC₅₀ values were not significant ($P > 0.05$), though mortality response increased with increase in exposure time and concentration of the test heavy metals.

Time	LC ₅	LC ₅₀	LC ₉₅	Toxicity factor
24h	66.633	105.213	166.130	1.71
48h	54.993	84.516	129.888	1.38
72h	46.286	71.424	110.214	1.16
96h	44.134	61.444	85.542	1.00

Table 5: Acute toxicity of binary mixture of Cd and Cr against *Clarias gariepinus* (mg/L).

Binary Acute Toxicity of Chromium (Cr) and Lead (Pb) against *Clarias gariepinus*

The LC₅₀ values for the exposure duration of *Clarias gariepinus* to the binary mixture of Cr and Pb are shown

Time	LC ₅	LC ₅₀	LC ₉₅	Toxicity factor
24h	17.074	54.646	174.904	2.08
48h	12.115	35.012	101.185	1.33
72h	12.023	32.172	86.088	1.22
96h	9.042	26.263	76.284	1.00

Table 6: Acute Toxicity of Binary Mixture of Cr and Pb against *Clarias gariepinus* (mg/L).

Joint Action Toxicity of Multiple Mixtures of Heavy Metal Compounds against *Clarias gariepinus*

The LC₅₀ values of multiple mixtures of Cd, Pb and Cr over 24- 96-h exposure duration are shown on Table 7. All the LC₅₀ values were statistically significant (P<0.05) except value of 24h duration (P>0.05) at 95% confidence limit.

Time	LC ₅	LC ₅₀	LC ₉₅	Toxicity factor
24h	32.537	70.893	154.465	3.97
48h	6.584	50.294	384.165	2.81
72h	6.003	31.828	168.739	1.78
96h	2.356	17.878	135.658	1.00

Table 7: Acute toxicity of multiple mixture of Cd: Pb: Cr against *Clarias gariepinus* (mg/L).

Bioaccumulation of Heavy Metals (Cd, Pb And Cr) By *Clarias gariepinus* Exposed to Sublethal Concentrations of Each Metal in Single Action Laboratory Studies

The heavy metal concentrations of single accumulation of the metal by *Clarias gariepinus* are shown in Table 8. The concentrations were derived based on the 1/10th and 1/100th, of their respective 96-h LC₅₀ values.

Days	Cd		Pb		Cr	
	1/10 th	1/100 th	1/10 th	1/100 th	1/10 th	1/100 th
0	ND	ND	ND	ND	ND	ND
7	2.216+0.001	0.272+0.001	0.118+0.001	0.018+0.001	89.630+0.001	9.752+0.001
14	2.503+0.001	0.311+0.001	0.121+0.001	0.021+0.001	106.410+0.001	10.281+0.001
21	2.904+0.001	0.337+0.001	0.125+0.001	0.022+0.001	113.920+0.001	10.810+0.001
28	3.079+0.001	0.351+0.001	0.128+0.001	0.022+0.001	131.812+0.001	12.544+0.001

Table 8: Bioaccumulation Concentrations of Cd, Pb and Cr in *Clarias gariepinus*(µg/g).

From the above table, there was increasing concentration of the metals in fish with increase in exposure duration for both 1/10th and 1/100th respectively. Chromium (Cr) was the metal mostly accumulated and Lead (Pb) was the least accumulated metal.

Bioaccumulation of Binary Mixture of Heavy Metals by *Clarias Gariepinus* Exposed to Sublethal Concentrations

The concentration of binary combination of Cd: Pb, Cd: Cr and Cr: Pb is presented in Table 9.

Days	Cd:Pb		Cd:Cr		Cr:Pb	
	1/10 th	1/100 th	1/10 th	1/100 th	1/10 th	1/100 th
0	ND	ND	ND	ND	ND	ND
7	1.811+0.001: 0.029+0.001	0.210+0.001: 0.002+0.001	2.331+0.001: 80.142+0.001	0.271+0.001: 8.560+0.001	29.191+0.001: 0.031+0.001	3.014+0.001: 0.002+0.001
14	1.873+0.001: 0.032+0.001	0.244+0.001: 0.002+0.001	2.416+0.001: 88.024+0.001	0.279+0.001: 9.112+0.001	30.013+0.001: 0.041+0.001	3.318+0.001: 0.003+0.001
21	1.980+0.001: 0.038+0.001	0.283+0.001: 0.003+0.001	2.782+0.001: 97.371+0.001	0.289+0.001: 9.976+0.001	30.834+0.001: 0.044+0.001	3.477+0.001: 0.003+0.001
28	2.325+0.001: 0.038+0.001	0.301+0.001: 0.003+0.001	2.814+0.001: 105.306+0.001	0.293+0.001: 10.423+0.001	32.181+0.001: 0.047+0.001	3.492+0.001: 0.004+0.001

Table 9: Bioaccumulation Concentrations of Cd: Pb, Cd: Cr and Cr: Pb in *Clarias gariepinus* ($\mu\text{g/g}$).

There was observed time-dose response relationship in the data sets in increasing direction with respect to the concentration of the metals in the exposed fish. Chromium (Cr) was also observed to be the metal with the highest concentration and Lead (Pb) the metal with the least concentration in the test animals.

Bioaccumulation of Equitoxic Multiple Mixtures of Heavy Metals (Cd: Cr: Pb)

The triple mixture of Cd: Cr: Pb concentrations are shown in Table 10.

Days	Cd:Cr:Pb	
	1/10 th	1/100 th
0	ND	ND
7	0.712+0.001:14.124+0.001:0.011+0.001	0.053+0.001:1.285+0.001:0.001+0.001
14	0.719+0.001:16.076+0.001:0.018+0.001	0.067+0.001:1.291+0.001:0.001+0.001
21	0.734+0.001:16.537+0.001:0.029+0.001	0.069+0.001:1.376+0.001:0.001+0.001
28	0.761+0.001:16.613+0.001:0.041+0.001	0.082+0.001:1.438+0.001:0.002+0.001

On the basis of the result, similar trend of dose response relationship was seen dependent on the exposure duration. The test animal bio accumulated chromium more than any other metal.

Discussion

Acute Toxicity Studies / Bioassay

Under the single action toxicity test carried out, Cd was the most toxic to the test animal with the lowest 96-h LC₅₀ of (8.280mg/L) (Table 4.7) while lead (Pb) was the least toxic (96-h LC₅₀ value of 70.183mg/L). The high toxicity of Cd has been previously demonstrated by several workers against fish species [5,6]. The high toxicity of Cd could be attributed to its high electropositivity. The low toxicity of Pb has also been reported widely in the literature. For example, Paul, et al. [7] was unable to establish a 96-h LC₅₀ value for lead acetate (90.43mg/L) against *Channa punctatus* because the compound did not cause any mortality of the exposed animal in their study "Lead toxicity on non-specific immune mechanisms of freshwater fish *Channa punctatus*".

Certainly, a number of interactions in joint toxic action of pollutants results in synergism in which the toxicity of a pollutant may be enhanced several folds by the presence of another pollutant against organisms. For example, Enajekpo [8] demonstrated that toxicity of spent engine oil against *C.africanus* was increased by over 2000 folds by the presence of Nuran insecticides in joint action studies.

In the joint action toxicity evaluations with binary and multiple (triple) mixtures carried out on this study, most of the results upon classification with the synergistic ratio agreed mainly with the model of synergism where only a few cases Cd in Cd: Pb, Cd in Cd: Cr and Cd in Cd: Pb: Cr interactions remained consistent with the model of antagonism. Liu, et al. [9] reported that the mechanism responsible for antagonistic interaction between constituent metal components in a mixture can be attributed to the oxidative stress and competition for uptake/binding sites in the biological interface between the various types of metals. Antagonistic interaction in mixture of pollutants Cd: Pb, Cd: Cr and Cd: Pb: Cr in this study, where it exists could be an advantage in

environmental management. This is so because antagonism implies that there is interaction between the constituents which result in the lowering of the toxicity of one or all the constituent of mixtures against the living species. Researchers outside the country who have also reported on synergistic interaction between constituent metals of a mixture include Abdullah and King [10] who documented the synergism that occurred in the interactions between Cu and Zn against freshwater isopod *Asellus aquaticus*. The observed synergistic interactions could also have occurred if the metals form complexes, which have greater penetrability with respect to the tissues of the exposed animal than the individual metals when acting alone

On the basis of emerging results obtained from the joint action toxicity studies carried out in this work, it will certainly be more realistic and effective to take into consideration toxicity levels of mixture of pollutants that occurs together in a local ecosystem and not just levels of single action toxicity of individual pollutants in deriving safe limits/standard aimed at protecting organisms in the environment. Although most of the existing safe limit/standards worldwide depended mainly on the existing single action toxicity data, view of such standards in the developed countries consider the results of joint action toxicity measurements against several sensitive species [11].

Bioaccumulation Studies

The bioaccumulation of Cd and Cr in the series of test carried out in this work likely occurred because the experimental animals were able to absorb the metals directly across body surfaces, membranes and ingested food at a faster rate than they were able to metabolize and excrete the absorbed metals. This led to a net gain of the metals in their body tissues. This is in line with the findings of Otitoloju and Don-Pedro [12,13] in their work on bioaccumulation of heavy metals (Zn, Pb, Cu and Cd) by *Tympanotonus fuscatus* exposed to sublethal concentrations of the test metal compounds in laboratory bioassays which recorded higher concentrations of metals that were about 2-6 times higher than the levels accumulated in control animals exposed to 30days period.

Another segment of this work reported here demonstrated the influence of binary and multiple mixtures of metals on the rate and pattern of bioaccumulation in the laboratory bioassays. When *Clarias gariepinus* were exposed to joint mixtures of Cd-Pb, Cd-Cr, Cr-Pb and Cd-Cr-Pb, in laboratory bioassays,

the concentrations of the metals accumulated by the test animals over 28-day experiments were found to be lower than the concentrations of the respective metals accumulated under the single action studies. This is in accordance with works of Borgmann, et al. [14] which stated that bioaccumulation of metals in mixtures may demonstrate competitive, a, or non-competitive inhibition that may lead to lowered concentrations when compared to single metal bioaccumulation.

Conclusion

During the single action acute toxicity studies, Cd was the most toxic heavy metal tested against *Clarias gariepinus* followed by Cr and Pb in a decreasing order of toxicity. There were significant departures when the toxicity levels of mixtures (binary and multiple) were compared to the toxicity levels of the individual metals when acting alone against the same test animal. *Clarias gariepinus* was found to bioaccumulate heavy metals (Cd, Cr and Pb) to varying degrees, dependent on the type of metals, period of exposure, and concentration of metal compound in the test media and the joint action of the metals in the system. Exposure of the test animal *Clarias gariepinus* to sublethal concentration of the metal mixtures under the joint action studies resulted in a reduction in the concentration of Cd, Pb, and Cr accumulated by the test organism, when compared to the concentrations accumulated by the animal during the single action studies.

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