



# Features of Energy of Chemical Reactions under the Action of Non-Lethal Acoustic Weapons

Fadeev GN\*, Boldyrev VS and Bogatov NA

Moscow State Technical University, Russia

\*Corresponding author: Fadeev GN, Moscow State Technical University, N Uh Bauman, Russia, Email: gerfad@mail.ru

Letter to Editor

Volume 2 Issue 1

Received Date: August 31, 2020

Published Date: September 22, 2020

## Abstract

The main features of energy of chemical reactions at low-frequency acoustic influences are experimentally defined in researches of scientific group of MSTU. N. Bauman in 2013-2017. Based on known experimental data (Table 1), attention was focused on the frequency range, including infrasound (up to 17 Hz) and the beginning of the sound range (up to 50 Hz). In experimentally established fundamental differences between the results of the influence of low-frequency acoustic oscillations of the action of ultrasound.

Frequencies, Hz	15-Feb	10-Feb	7-Jun	10-May
Intensity of Frequencies, decibels	95-105	125	90	135
Reaction organism's	Slowdown visual reactions	Unpleasant feelings internal bodies'	Violation the alpha-rhythm brain	Rapid heartbeat

**Table 1:** Influence of acoustic vibrations on the human body.

## Experimental Result

20-22 Hz.

The study of clathrates of iodine [1,2] revealed the *extreme nature of the effect* of infrasonic oscillations. It is established that the action of low acoustic frequencies (from 1 to 50 Hz) is determined by *the optimal frequency at which the speed of a particular sound chemical transformation is maximal*. To determine whether the detected sensitivity to low-frequency effects is a feature of individual complexes or it is inherent in all clathrate structures, special experiments were conducted. It turned out that even the components of starch – amylose and amylopectin, which are close in composition, are sensitive to different frequencies in the formation of the corresponding clathrate complexes with iodine [2]:

- Amyloidin- to infrasonic frequencies 12-15 Hz,
- Amylopectin- to the beginning frequency sound range

On the basis of the conducted experiments, it can be concluded: biochemical systems sensitive to the external infrasound effect, having clathrate structure, can serve as a basis for the development of antidotes against non-lethal acoustic weapons [3-5]. It was interesting to find out whether this behavior is related to the special properties of clathrates, or sensitivity to low-frequency effects is inherent in other biochemical active structures. As biochemically active models that have a chelate structure, different from the clathrate were taken from the polyvinylpyrrolidone-iodine. Destabilization of this compound in the field of low-frequency effects [6-8] proves that sensitivity to the action of low-frequency oscillations is a property of a significant number of biochemical active structures. This fact, established by the conducted research, indicates the principal directions of creation of antidotes against non-lethal acoustic weapons.

## Discussion of Results

Experimental researches have allowed to establish the main: *can be active only part of the energy of the JAC of acoustic waves, which is absorbed by the system* (similar to the law of Grothus - Draper in photochemistry). The rest of the energy pulsating with low-frequency acoustic effects of bubbles under the condition of "undeveloped" cavitation is consumed:

- Local electrification of bubbles,
- The creation of the cavitation surge,
- The formation of cumulative jets.

As a consequence, in energy and kinetic calculations should not take into account all the acoustic energy introduced into the solution, but only its share spent on the formation of ions, free radicals or restructuring of the structure of molecules (for example, the translation of dimers into monomers or back) and other processes. This energy is called the chemical-acoustic energy  $E_{ca}$ . Its connection with the total acoustic energy of the  $E_{ac}$  was proposed by L. D. Rosenberg [9] to estimate via the constant  $\eta_{ca}$ , which, in essence, is the efficiency of the sound chemical process:

$$E_{ca} = \eta_{ca} \cdot E_{ac}$$

Where  $\eta_{ca}$  is chemical-acoustic efficiency. In the study of physical and chemical processes occurring in the cavitation bubble and the adjacent volume of liquid, it is assumed that the energy released in the process of acoustic effects is sufficient to excite, ionization and dissociation of water molecules, gases and substances with high pressure of saturated steam inside the cavitation bubble. At this stage, the action of acoustic oscillations on substances penetrating into the cavity is direct, i.e. *direct action*.

The so-called *indirect action* of acoustic oscillations is carried out after the transition to a solution of active particles that have arisen in the system. They react with dissolved substances, solvate with solvent molecules or move to a different state than the original. Thus, the reaction of particles formed during sonolysis can be divided into two types:

- Recombination in the cavitation bubble;
- Reactions in the volume of the solution.

To assess turning Margulis MA, et al. [10] introduced the concept of elementary chemical acoustic output FR. The value of FR is equal to the number of particles of sonolysis products formed for every 100 eV of the spent chemical and acoustic energy and transferred to the solution (without taking into account their further reactions in the solution). To account for further transformations, the concept of  $F_{R2}$  -

secondary energy yield of chemical reactions (or chemical-acoustic output  $F_{R2}$ ) was introduced. This value is equal to the number of particles formed (or disappeared) for every 100 eV of absorbed chemical and acoustic energy.

The fundamental characteristic of acoustic energy (as well as other types of radiation) is the *recombination coefficient*  $\beta_R$  - the ratio of chemical-acoustic outputs of recombinant particles  $F_{R2}$  to the value of  $F_R$  particles before recombination:

$$\beta_R = F_{R2} / F_R$$

The value of  $\beta_R$  - depends on many factors: the density or concentration of energy, the size of the impact area and its localization, the number and concentration of the resulting particles and other factors. Given the General nature of the recombination coefficient  $\beta_R$ , it can be used to compare the acoustic energy with other types of physical effects on the studied structures.

## References

1. Fadeev GN, Boldyrev VS, Tveritinov VN, Pashkova LI (2013) Clathrates of iodine-prototypes of antidotes against acoustic weapons. Vestnik MGTU, Series Natural Sciences 1(44): 82-88.
2. Fadeev GN, Boldyrev VS, Ermolaeva VI (2013) Biologically active clathrates, amiloidin, and amilopektoiodin under exposure to low-frequency. 7<sup>th</sup> European Symposium on Non-Lethal Weapons (Ettlingen, Fraunhofer-Institut fur Chemische Technologie), Germany, pp: 50-1-50-8.
3. Boldyrev VS (2013) Biochemical active structures in the field of low-frequency acoustic effects. Diss kand tech sciences 127.
4. Fadeev GN, Boldyrev VS, Ermolaev I (2014) The physics systems in the field of low-frequency effects. Physical and Mathematical Problems of Advanced Technology Development, Abstracts, Moscow, pp: 93-94.
5. Fadeev GN (2015) Biochemical active systems-antidotes against low-frequency influences. SB. of reports 2<sup>nd</sup> International scientific-practical conference "non-lethal weapons non-lethal actions Moscow, Advertising Agency, Aleksv, pp: 50-51.
6. Fadeev GN, Boldyrev VS, Sinkevich VV (2015) Sonochemical transformations of chelate and clathrate structures in a low-frequency acoustic field. Doklady Physical Chemistry 462(2): 119-121.
7. Fadeev GN, Ermolaev VI, Boldyrev VS, Sinkevich VV (2016) Kinetics of destabilization of polyvinylpyrrolidone-

- iodine in the field low-frequency effects. Journal of physical chemistry 90(9): 289-292.
8. Boldyrev VS, Ermolaev VI, Sinkevich VV, Fadeev GN (2017) Destabilization of polyvinylpyrrolidone-iodine in the field low-frequency effects. Journal Vestnik MSTU, Series Natural science 4: 90-99.
  9. Rozenberg LD (1965) The cavitation On the evaluation of the effectiveness acoustic energy. Acoustic Zh 11(1): 121-124.
  10. Margulis M, Maltsev AN (1968) On the evaluation of the energy yield of chemical reactions initiated by ultrasound. I. Chemical-acoustic output reactions. Journal of physical chemistry 42(6): 1441-1446.

