



Use of 9.3 μ m CO₂ Laser for Removal of Zirconia Restorations

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Abstract

Objective: The primary aim of this study was to test if a 9.3 μ m Carbon Dioxide (CO₂) laser could be used to effectively separate a cemented zirconia restoration from dentin. The secondary aim was to compare shear bond strength (SBS) between RelyX™ Unicem 2 Automix Cement (3M ESPE) (RelyX™ Unicem 2) and RelyX™ Luting Plus Automix Cement (3M ESPE) (RelyX™ Luting Plus).

Methods: 40 teeth were prepared to expose dentin and then randomized into four groups of 10 samples. Zirconia slices (non-glazed Yttrium stabilized zirconia samples (2.5 x 3 x 1.5 mm)) were sandblasted with 50 micron aluminum oxide at 30 psi and then cemented onto the dentin samples with RelyX™ Unicem 2 or RelyX™ Luting Plus, dependent on the group. The cements were applied to both the zirconia and dentin with a force of 20 g/mm² for 30 seconds following manufacturer's recommendations. After 48 hours, the 9.3 μ m CO₂ laser was used on half the samples for 5 seconds. All 40 zirconia samples were removed with the Instron 5566A in a traditional SBS test. The groups were: 1A - RelyX™ Unicem 2 and Laser, 1B - RelyX™ Unicem 2 and Shear Bond, 2A - RelyX™ Luting Plus and Laser, and 2B - RelyX™ Luting Plus and Shear Bond. The Mann-Whitney U test was used for comparison.

Results: Mean SBS of four groups ranged from 0.5 to 4.4 MPa. There was a significant difference in the SBS between 9.3 μ m CO₂ laser and Shear Force methods for RelyX™ Luting Plus. However, the difference between the two methods was not significant for RelyX Unicem 2. RelyX™ Unicem 2 provided significantly higher SBS than RelyX™ Luting Plus for both the CO₂ laser and shear force methods (Table 2).

Conclusion: The 9.3 μ m CO₂ laser effectively separated the zirconia restoration cemented with RelyX™ Luting Plus from dentin. RelyX™ Unicem 2 provided significantly stronger SBS than RelyX™ Luting Plus.

Keywords: Zirconia; Laser; Shear Bond Strength; Zirconia Removal

Abbreviations: SBS: Shear Bond Strength; CO₂: Carbon Dioxide; Er:YAG: Erbium-doped Yttrium Aluminium Garnet; Nd:YAG: Neodymium Yttrium Aluminum Garnet.

Introduction

Crown placement involves preparation of the tooth, removing enamel, creation of a crown, adjusting of the crown, and its cementation to the remaining tooth structure [1]. It is important to use restorative materials that can withstand large amounts of occlusal forces. Due to its strength and

esthetic properties, zirconia is becoming widely used in dentistry for restorations including crowns [2]. However, when crowns need to be replaced due to development of carious lesions or fractures, the crown removal process is time consuming. It typically includes using diamond burs to cut the zirconia crown with the use of a high-speed handpiece at 150,000 rpm and a 0.9 N cutting force [2]. This process is followed by torquing the edges to shear the crown.

Several lasers have been developed over the last couple of decades in dentistry [3,4]. Such lasers include CO₂,

Neodymium Yttrium Aluminum Garnet (Nd:YAG), and Er:YAG and can be used on soft or hard tissues. Soft tissue laser uses include healing of injured tissue, exposing impacted or partially erupted teeth, and cancer therapies [4]. Laser usage on hard tissue includes removing restorative material, bleaching, preventing tooth sensitivity, and preventing cavities [4]. Additionally, the CO₂ laser has been successful in increasing caries resistance of enamel [5-8].

Important factors for selecting a dental technique are patient comfort, chair time, and cost. Concerning crown removal, it is time consuming to use burs and handpiece instruments, which additionally generate heat [3]. Hard tissue lasers provide an alternative form of ceramic crown removal. Reducing the number of burs will decrease the cost of constant bur replacement during crown removal. Another advantage of using a laser is the reduction of the noises and vibrations experienced with a high-speed instrument, and the associated discomfort experienced by the patient [3]. Lasers prove to be successful in increasing efficiency and comfort while decreasing costs of dental procedures.

Many dentists and researchers alike are looking for possible alternatives to using burs and shear force to remove crown restorations. One study was conducted using an Erbium-doped Yttrium Aluminium Garnet (Er:YAG) laser to debond and remove the crown from the tooth structure. Both zirconia and lithium disilicate crowns were removed using this laser with no fractures or damage to the tooth structure or the crown [9]. The ability of the Er:YAG laser on crown removal leads to the potential use of another laser, the CO₂ laser, for the same purpose.

The Er:YAG laser has shown to be successful in eliminating cement from the tooth structure [4]. Due to its affinity for water, it is also useful in hard tissue treatment [4]. Likewise, the CO₂ laser has an affinity for water [4]. Since the two lasers share this water affinity characteristic, the removal of cement and restorations which has been confirmed with the use of the Er:YAG laser may also be possible with the

use of the CO₂ laser. Cemented zirconia restorations can be very challenging to remove, as the bonding can be very strong. Cranska reported the crown was removed within one minute with the use of an Er:YAG laser [3]. An Er:YAG laser is effective in removing all-ceramic restorations including zirconia crowns [3,9].

Er:YAG lasers, along with CO₂ lasers have been used in the dental industry for surface treatment of restorations. Er:YAG lasers were found successful at increasing shear bond strength (SBS) after surface treatment [11]. 9.3µm CO₂ lasers also increased SBS after surface treatment [10-15]. Based on the concept that the Er:YAG laser can successfully debond the crown from the tooth, it was hypothesized that a 9.3µm CO₂ laser could also separate zirconia from tooth structure.

The aims of the study were primarily to determine if a 9.3µm CO₂ laser could separate cemented zirconia from underlying tooth structure and secondarily to compare SBS between cements.

Materials and Methods

A total of 40 blocks of zirconia were used in this study. Molar and premolar teeth were collected from oral surgery offices in the greater Chicago area and stored in a solution of 10% Clorox bleach in water until used. 40 molars and premolars were embedded into cylindrical acrylic resin blocks and sliced with the IsoMet saw at 850 rpm to expose the dentin of each sample. The non-glazed Yttrium stabilized zirconia samples (2.5 mm x 3 mm x 1.5 mm (length x width x thickness)) were obtained from Stanford Advanced Materials. The zirconia samples as well as the dentin samples were randomized into four groups, namely Group 1A (RelyXTM Unicem 2, Laser), Group 1B (RelyXTM Unicem 2, Shear Bond), Group 2A (RelyXTM Luting Plus, Laser), Group 2B (RelyXTM Luting Plus, Shear Bond) with 10 samples in each group. Each zirconia block was cemented to the dentin tooth slice with the cement outlined in Table 1 (as it mimics zirconia crown restorations).

	Group 1A	Group 1B	Group 2A	Group 2B
Cement	RelyX™ Unicem 2	RelyX™ Unicem 2	RelyX™ Luting Plus	RelyX™ Luting Plus
Treatment	Laser	Shear bond strength	Laser	Shear bond strength

Table 1: Group Outline and Conditions.

Sample Preparation

All zirconia samples were sandblasted with 50 micron aluminum oxide at a pressure of 30 psi, cleaned with alcohol, and air dried. All cements were applied to both the zirconia and dentin and cemented with a constant force of

20 g/mm² for 30 seconds, following the manufacturer's recommendations. More specifically, for Group 1A and 1B with RelyX™ Unicem 2, the samples were tack cured with a Demi™ Plus Curing Light for 2 seconds, excess cement was removed with an explorer while holding the slices together, then light cured for another 20 seconds. Any excess cement

was removed after an additional 6 minutes. For Group 2A and 2B with RelyX™ Luting Plus, the working time of the cement is 1.5 minutes according to the 3M manufacturer's recommendations. The samples were tack cured with a Demi™ Plus Curing Light for 5 seconds, excess cement was removed with an explorer while the slices were held together. The cement was left to set for an additional 5 minutes prior to removing any other cement excess with an explorer.

The cemented samples were stored in water for 48 hours. As recommended by the manufacturer, the 9.3µm CO₂ laser (Solea, Convergent Dental, Needham, MA) was set to a 1 mm spot size with an average power of 23.6 W. Laser irradiation was performed for 5 seconds continuously on each sample positioned at the laser's focal point, which is 10 mm from the tip of the handpiece (Figure 1). The 9.3µm CO₂ laser was used only on samples of Group 1A and 2A. Zirconia and dentin separation was determined by visual inspection of each sample. For those samples that the zirconia was not separated with the use of the CO₂ laser, the SBS was tested using the Instron 5566A model at 1 mm/minute crosshead speed and recorded. For group 1B and 2B samples, the SBS was tested using the Instron 5566A model at 1 mm/minute crosshead speed and recorded for each sample without treatment of the CO₂ laser. This machine is a universal testing device. One of its many tests is cements' adhesive ability. This study used the Instron 5566A to test the adhesion between the zirconia ceramic and the tooth structure. SBS is important to demonstrate the force needed to separate the zirconia.

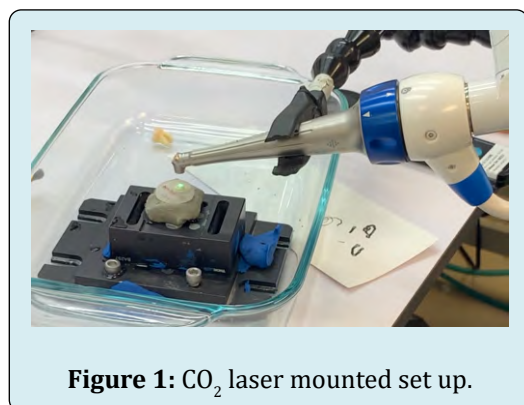


Figure 1: CO₂ laser mounted set up.

Sample Size

The sample size calculation was performed using nQuery Advisor. Since literature on CO₂ lasers was sparse and could not be used for calculating sample size, the sample size was calculated based on the secondary aim (comparing SBS of RelyX™ Unicem 2 and RelyX™ Luting Plus) of the study. Based on the results of Blatz MB, et al. the mean±SD SBS of RelyX™ Luting Plus and RelyX™ Unicem 2 were assumed to be 5.75±1.53 MPa and 10.95±4.28 MPa respectively [16]. Based on these assumptions, a sample size of n=10 samples per group would provide 90% power to detect an effect size of 1.618 with a Type I error rate of 0.05.

Statistical Analysis

Means, standard deviations, medians and interquartile ranges were calculated for SBS. Normality of SBS data was assessed graphically and using the Shapiro-Wilk test. The Mann Whitney U test was used to compare SBS between the laser and shear force method for each cement. It was also used to compare SBS between RelyX Unicem and RelyX Luting for each method. Stata version 16 was used for analyses and significant value was set at 0.05.

Results

The mean SBS and standard deviations of each of the four groups are presented in Table 2, ranging from 0.5 to 4.4 MPa. 1 of the 10 samples in the RelyX™ Unicem 2 laser group and 5 of the 10 samples in the RelyX™ Luting Plus laser group successfully separated with the use of the laser only and did not undergo the SBS test with the Instron 5566A. It was determined that there was a statistically significant difference in SBS between the use of the laser and shear force method for RelyX™ Luting Plus, with a p-value of 0.01. However, the difference was not statistically significant between the two methods for RelyX™ Unicem 2, with a p-value of 0.20. SBS of RelyX™ Unicem 2 was significantly higher than RelyX™ Luting Plus for both the laser method and shear force method, with p-values of 0.04 and 0.03 respectively. Figure 2 is a 3D rendering of the zirconia block after the CO₂ laser was used on the zirconia sample.

	RelyX™ Unicem 2	RelyX™ Luting Plus	p**
CO ₂ Laser	3.4 ± 4.3	0.5 ± 0.9	0.04
Shear Force	4.4 ± 2.4	2.0 ± 2.0	0.03
p*	0.2	0.01	

Data was presented as mean and standard deviation.

p*: p-value for comparison between methods (primary aim)

p**: p-value for comparison between cements (secondary aim)

Table 2: Shear bond strength (SBS) in MPa of four groups.

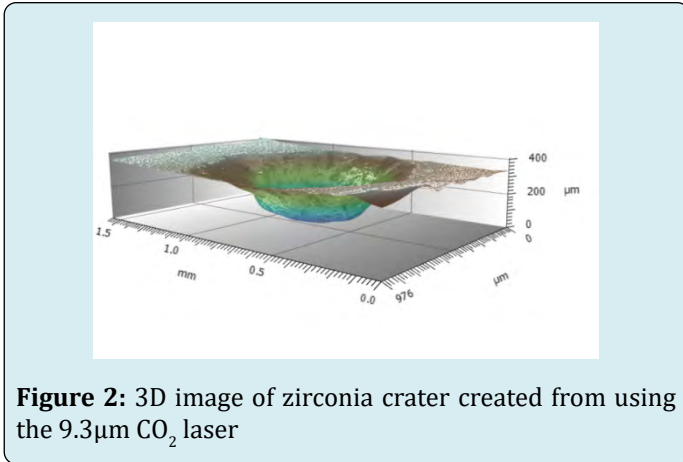


Figure 2: 3D image of zirconia crater created from using the 9.3µm CO₂ laser

Discussion & Conclusion

The primary aim of this study was to identify the correlation between the use of a CO₂ laser to the use of shear force in assessing the separation of zirconia blocks cemented to dentin. The results from this study suggest that the CO₂ laser can be used to remove zirconia from underlying tooth structure when cemented with RelyXTM Luting Plus.

The secondary aim of this study was to compare the SBS of the cements. The results were conclusive with previous literature that RelyXTM Unicem 2 has a higher SBS than RelyXTM Luting Plus [16,17], which may contribute to the difficulty in removing zirconia from dentin with the laser when cemented with RelyXTM Unicem 2 compared to RelyXTM Luting Plus. This may suggest that a longer laser time, higher laser pattern, or modified pulse rate might be needed in order to successfully remove the zirconia from dentin when cemented with RelyXTM Unicem 2.

It is also important to note the solubility and absorption of the two cements. RelyXTM Luting Plus is a resin modified glass ionomer cement, while RelyXTM Unicem 2 is a self-adhesive universal resin cement. RelyXTM Unicem 2 [18,19] has a lower solubility and lower sorption than RelyXTM Luting Plus [20]. Higher solubility and water sorption signifies higher water content. The water content in the cement affects the laser energy necessary to separate the zirconia from tooth structure. The higher the water content in the cement, a decreased amount of laser energy should be needed to debond the two surfaces [4]. Er:YAG laser energy is mostly absorbed in water [4,5,9,11]. The CO₂ laser also has a very large affinity for water [4]. This could yield in less water activation from the laser with the RelyXTM Unicem 2 samples, causing it to be more difficult to separate the zirconia from dentin.

There are currently two theories for the absorption process. The free volume approach hypothesizes that water

molecules disperse in the voids of the resin in which they interact with, which may cause hydrolytic degradation [19]. Another theory suggests bonding of water and the cement's hydrophilic groups, such as phosphate groups, causing hygroscopic expansion [19].

Although both the 9.3µm CO₂ laser and the Er:YAG laser have a high affinity for water, the Er:YAG laser's emission wavelength is located at the peak of water absorption, resulting in a 10 to 15 fold stronger affinity for water than the CO₂ laser [21]. On the other hand, the CO₂ laser emission wavelength's absorption is predominantly in ceramics [11-13,15]. Because of this, the 9.3µm CO₂ laser can cause porosities in the zirconia [11,13], which in turn may weaken the bond between the zirconia and cement, instead of weakening the cement itself through the cement's water content.

In practice, it is beneficial to dentists to have additional ways to complete a procedure. Many dentists already use 9.3µm CO₂ lasers for surface treatment. This study determined that 9.3µm CO₂ lasers could be used to separate zirconia from the tooth structure when cemented with RelyXTM Luting. Based on the results of this study, additional testing will need to be performed to confirm if this bench method could be used clinically. Next steps to consider include using zirconia crowns instead of blocks and using a thermocouple at the interface of the zirconia and tooth structure to measure the heat directed from the laser to the tooth. The zirconia blocks used were 1.5 mm thick to resemble the thickness of a standard zirconia crown. However, when the CO₂ laser will be tested on zirconia crowns, modifications may include a higher energy pattern, longer laser time, or other laser settings to successfully remove the zirconia crown from dentin.

Within the limitations of this study, the 9.3µm CO₂ laser effectively separated the zirconia restoration cemented with RelyXTM Luting Plus from dentin. RelyXTM Unicem 2 provided significantly stronger SBS than RelyXTM Luting Plus.

Conflict of Interest

This study was sponsored in part by Convergent Dental.

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