

# Transferosomes for Trans-Nasal Brain Delivery of Clonazepam: Preparation, Optimization, *Ex-Vivo* Cytotoxicity and Pharmacodynamic Study

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## Abstract

Clonazepam (CZ) is a benzodiazepine derivative that is used mainly in the treatment of status epilepticus (SE). Intranasal (IN) administration of nano-carrier systems generally have the ability to deliver the drug to the brain through olfactory and trigeminal nerves pathways circumventing the blood brain barrier (BBB). The aim of the present work was to enhance intranasal (IN) brain delivery of clonazepam in an attempt to treat status epilepticus (SE). In order to achieve this goal, clonazepam transferosomes (TF) were prepared *via* thin film hydration technique (TFH) adopting 2<sup>2</sup> X 3<sup>1</sup> full factorial design (FFD). The twelve formulae were evaluated in terms of entrapment efficiency (EE), particle size (PS), polydispersity index (PDI), zeta potential (ZP) and *in-vitro* release. TF1 prepared using sodium deoxycholate (SDC) as edge activator (EA), lipid to edge activator molar ratio of 10:1 and 10 mg as initial drug amount showed the highest desirability value of 0.833. The optimized formula showed minor changes to sheep nasal mucosa upon *ex-vivo* cytotoxicity study and significant delay in the onset of pentylenetetrazole (PTZ) induced seizures. The declared results reveal the ability of the developed TF to be a strong potential candidate for the emergency treatment of SE.

**Keywords:** Intranasal; Transferosomes; Clonazepam; Cytotoxicity; Pentylenetetrazole.

**Abbreviations:** TEM-Transmission Electron Microscopy; BBB-Blood Brain Barrier; CMC-Critical Micelle Concentration; DSC-Differential Scanning Calorimetry; EE-Entrapment Efficiency

## Introduction

Epilepsy represents a group of brain disorders which have in common the occurrence of spontaneous and recurring epileptic seizures i.e., hyper synchronous electrical discharge in brain network [1]. Status epilepticus (SE) is a medical emergency that is associated

with high morbidity and mortality [2]. It is defined as a prolonged seizure or multiple seizures with incomplete return to baseline [3]. Neuronal damage in SE results from sustained NMDA-mediated neuronal stimulation which leads to apoptosis [4]. When these neuronal cells are depolarized, the Mg<sup>2+</sup> ions diffuses outward, allowing sodium ions and Ca<sup>2+</sup> to flood the cell, resulting in a cascade of Ca<sup>2+</sup>-mediated cytotoxic events, leading to neuronal injury, cell lysis, and cell death [5]. Treatment of SE should aim to immediately terminate the seizures with prevention of their recurrence [6].

Clonazepam (CZ) is a nitro benzodiazepine derivative. Its structure is 5-(2-Chlorophenyl)-7-nitro-2,3-dihydro-1,4-benzodiazepin-2-one. It acts *via* modulating gamma-aminobutyric acid (GABA) function in the brain, *via* the benzodiazepine receptor which in turn leads to enhanced GABA ergic inhibition of neuronal firing. In addition clonazepam decreases the utilization of 5-HT (serotonin) by neurons and has been shown to bind tightly to central type benzodiazepine receptors [7]. It is said to be among the class of “highly potent” benzodiazepines. Clonazepam is available as tablet dosage forms, with peak plasma concentration between 1 and 4 hours after ingestion [8]. It is extensively metabolised in the liver, its principal metabolite being 7-aminoclonazepam, which has no antiepileptic activity. It is about 85% bound to plasma proteins and estimations of its elimination half-life range from about 20 to 40 hours [9]. As a highly lipophilic molecule and highly protein bound, CZ is characterized by high systemic distribution in fat leading to adverse effects such as palpitation, hair loss and anorexia among others [10].

Drug delivery to the brain is a challenge because of the presence of the blood brain barrier (BBB). Intranasal administration of the neurotherapeutics results into direct delivery of the drug to the brain circumventing the BBB. This route is generally non-invasive, convenient, self-administered, along with rapid drug absorption leading to quick onset of action [11]. On the other hand, oral or intravenous drug delivery is inconvenient and impractical for the control of convulsions due to patient's inability to swallow and the need for qualified personnel for intravenous administration.

Transferosomes (TF) are highly adaptable and stress-responsive, complex aggregates [12]. Each transfersome consists of at least one inner aqueous compartment, which is surrounded by a lipid bilayer with specially tailored properties [13] and a bilayer softening component (such as a biocompatible surfactant) that increases lipid bilayer flexibility and permeability [14]. They have been used to target olanzapine directly to the brain after intranasal application according to the work of

Salama et al. [15]. The results demonstrated the ability of the developed nanocarriers for brain targeting of the drug.

Hence, the aim of the present study was to develop CZ loaded TF intended for brain delivery of the drug *via* the intranasal route in an attempt for immediate control of SE.

## Materials and Methods

### Materials

Clonazepam (CZ) was a gift from Amoun Pharmaceuticals ( Elobour city- Egypt), Sodium deoxycholate (SDC) was purchased from Sigma Chemical Co. (USA), Labrafil (LB) was a gift from Gattefosse (France), Phospholipon 90G was obtained as a gift from Lipiod GmbH (Ludwigshafen, Germany). Spectra/Pore® dialysis membrane (12,000–14,000 molecular weight cut off) was purchased from Spectrum Laboratories Inc. (CA, USA). Ethanol and methylene chloride were purchased from El-Nasr Chemical Co. (Cairo, Egypt).

### Experimental Design

In order to investigate the influence of formulation variables on the transferosomes characteristics, CZ loaded transferosomes were prepared according to full factorial design (FFD) ( $2^2 \times 3^1$ ) using Design Expert® software (Version 7, Stat-Ease Inc., Minneapolis, MN). The independent variables investigated were  $X_1$ : edge activator type (EA),  $X_2$ : lipid: edge activator molar ratio and  $X_3$ : initial drug amount. For  $X_1$  and  $X_3$ , the levels of each factor were designated as (-1, +1). Concerning  $X_2$ , the designated levels were (-1, 0, +1). The corresponding actual values of the independent variables are shown in Table 1. The compositions of the 12 formulae of the full factorial design are shown in Table 2. Analysis of variance (ANOVA) was carried out to estimate the significance of model and term. Probability p-values ( $p < 0.05$ ) denoted significance.

Factors	Levels		
	-1	0	1
$X_1$ :EA type	SDC		LB
$X_2$ :lipid : EA ratio	10:1	30:1	50:1
$X_3$ :initial drug amount	10		20

EA: edge activator, SDC: sodium deoxycholate, LB: labrafil

Table 1: The Independent Variables and their Levels for  $2^1 \times 3^1$  Factorial Design for CZ Loaded TF.

Formula code	Factor's levels in actual values		
	X <sub>1</sub>	X <sub>2</sub>	X <sub>3</sub>
TF1	SDC	10:1	10
TF2	SDC	30:1	10
TF3	SDC	50:1	10
TF4	SDC	10:1	20
TF5	SDC	30:1	20
TF6	SDC	50:1	20
TF7	LB	10:1	10
TF8	LB	30:1	10
TF9	LB	50:1	10
TF10	LB	10:1	20
TF11	LB	30:1	20
TF12	LB	50:1	20

Table 2: Composition of 2<sup>2</sup> X 3<sup>1</sup>FFD for CZ transferosomes

### Preparation of Clonazepam Loaded TF

TF were prepared adopting the thin film hydration technique [16]. Predetermined weights of CZ (10, 20 mg) and 300 mg mixture of phospholipon 90G and the EA (SDC or LB) according to the specified molar ratio were accurately weighed and dissolved in 10 ml methylene chloride in a one liter round-bottomed flask. Methylene chloride was slowly evaporated under vacuum at 50° C for 30 min. using rotary evaporator at 90 rpm such that a thin dry film of the components was formed on the inner wall of the flask. The dried thin film was hydrated with 10 ml distilled water by rotating the flask in water bath at 30° C by rotary evaporator with the aid of small glass beads (25 small glass beads, each of a diameter of 2mm) at 210 rpm for one hour under normal pressure. The prepared TF were then refrigerated (5°C±3) in amber colored vials for further investigation.

### In-Vitro Evaluation of the Prepared TF

**Determination of entrapment efficiency:** Methanol was selected as an appropriate solvent for the lysis of the prepared transferosomes [17]. Total amount of the drug (drug content) of the prepared formulae was determined by dissolving 0.5ml of the prepared TF in methanol and then measuring the UV absorbance using spectrophotometer at the predetermined  $\lambda_{\max}$  of CZ in methanol after performing the necessary dilution. In order to determine EE, CZ containing transferosomes were separated from the untrapped drug by filtrating the formed suspension using Whatman filter paper

(Grade No. 1, 11  $\mu\text{m}$ ) [18]. Owing to its extremely low solubility in water, the untrapped CZ was sedimented and retained over the filter paper. 0.5 ml of the separated vesicles (the filtrate) was disrupted by sonication with methanol and the concentration of the entrapped drug was measured spectrophotometrically at the same  $\lambda_{\max}$ . The EE was calculated using the following formula:

$$EE \% = \frac{\text{amount CZ entrapped (mg)}}{\text{total amount of CZ (mg)}} \times 100$$

The measurements were done in triplicates and the results were recorded.

**Determination of Particle Size (PS), Polydispersity Index (PDI) and Zeta Potential (ZP):** The mean PS, PDI and ZP were determined by Zetasizer (Malvern Instruments Ltd., Worcester-shire, UK) at 25 °C adopting the dynamic light-scattering method. The results were recorded in triplicates.

**Determination of Deformability Index:** Comparative measurement of elasticity was performed by extruding vesicles suspension through nylon millipore filters with a pore size of 200 nm [19] at a constant pressure of 2.5 bar [20]. Elasticity of the vesicle membrane was expressed in terms of deformability index (DI) according to the following equation [21]:

$$DI = J \left( \frac{r_v}{r_p} \right)$$

Where *DI* is the deformability index (g), *J* is the weight of dispersion extruded in 2 minutes, *r<sub>v</sub>* is the size of vesicles

after extrusion (nm), and  $rp$  is the pore size of the barrier (nm).

**In-vitro Release:** The CZ release from the developed TF was assessed in triplicates using the membrane diffusion technique [22]. Volumes of the filtered transferosomes containing 1mg of the drug were placed in a dialysis bag (soaked overnight in distilled water) immersed in 50 mls of the release medium (ethanol: water 1:1) [23]. The bottles were then shaken in a thermostatically controlled shaking water bath operating at 100 shake per minute and a temperature of  $37\text{ }^{\circ}\text{C}\pm 0.5$  [24]. At predetermined time intervals, three mls of the release media were withdrawn and immediately replaced by an equal volume of fresh medium. Drug amount was then assayed spectrophotometrically at  $\lambda_{\text{max}}$  309 nm. Percentage of drug released was calculated and plotted versus time. The same procedure was repeated using the drug solution (in ethanol/water mixture in the ratio of 1:1 and conc. of 1mg/2ml) to test the ability of the drug to cross the dialysis bag. The time required for the release of 50% of the loaded drug ( $t_{50\%}$ ) and dissolution efficiency (DE) were also calculated and statistically analyzed.

Kinetic analysis was subsequently done according to zero and first orders and Higuchi diffusion model [25]. The model with the highest coefficient of determination was considered the best fitting.

**Selection of the Optimized TF Formula:** Desirability was calculated using Design-Expert® software and considered to optimize the studied responses depending on the provided results. The significant responses were taken into considerations while the non-significant factors were not [26]. The TF formula with the highest desirability value (close to 1) was subjected for further investigation.

**Differential Scanning Calorimetry (DSC):** Samples (2mg) of pure drug, components (Phospholipon 90G and SDC) and drug loaded optimized formula (TF1) were heated in an aluminum pan at a rate of  $5^{\circ}\text{C}/\text{min}$  in an atmosphere of nitrogen to  $400^{\circ}\text{C}$  and the thermo grams were recorded (15).

**Transmission Electron Microscopy (TEM):** One drop of the optimized formula (TF1) was placed on a copper grid and the excess was removed using a filter paper and left to dry at room temperature. Then, one drop of 2% phospho-tungstic acid aqueous solution was added (negative staining, to improve contrast between the inorganic materials and the underlying supporting film)

and the excess was similarly removed and similarly dried. Finally, the grid was examined under a transmission electron microscope (Jeol JEM 2100, Tokyo, Japan).

**Effect of storage:** The investigated formula (TF1) was assessed following storage at refrigerator ( $5^{\circ}\text{C}\pm 3$ ) over four weeks. At the end of the storage period, transferosomes were evaluated with respect to their appearance, EE, PS, and Q 8h. Statistical analysis of the obtained results was performed by Student's-  $t$ -test using SPSS 17.0® software (Chicago, USA). Differences at  $p \leq 0.05$  were considered significant. The release profile of the stored transferosomes was compared to that of the freshly prepared ones according to the model independent mathematical approach of Moore and Flanner [27]. The similarity factor ( $f_2$ ) was calculated according to the following equation:

$$f_2 = 50 \log \left\{ \left[ 1 + \left( \frac{1}{n} \sum_{t=1}^n (R_t - T_t)^2 \right)^{-0.5} \right] \times 100 \right\}$$

where  $n$  is the number of sampling points,  $R_t$  and  $T_t$  are the mean percent released from reference (fresh) and from test (stored) at time  $t$ , respectively. An  $f_2$  value  $\geq 50$  indicates that the release profiles are similar, whereas smaller values may imply dissimilar release profiles.

### Ex-Vivo Assessment of Nasal Cytotoxicity of the Optimized Formula

The protocol of the study (PI 1114) was reviewed and approved by Research Ethics Committee-Faculty of Pharmacy, Cairo University (REC-FOPCU) in Egypt. Histopathological analysis was done on isolated sheep nasal mucosa to assess the possible local cytotoxic effects of the developed CZ loaded transferosomes. Segments from the anterior and posterior nasal mucosa were dissected from the head of a 1.3 old sheep weighing 55 kg obtained from the local slaughter house (Cairo-Egypt), within 10 minutes of the sacrifice. The nasal cavity was exposed longitudinally. The obtained tissues were immediately washed with normal saline. The excised tissues were then randomly allocated to three groups so that each group contains equal number of anterior and posterior segments. Group one was treated with pH 6.4 phosphate buffer saline (PBS) as negative control [28], group two received isopropyl alcohol as a positive control [29] and the third group was exposed to CZ-loaded TF, for two hours. All the three groups received equal volumes of the treatment (2ml). The pieces were washed with distilled water and preserved in 10% formalin in saline solution [30]. The mucosal tissues were stained with hematoxylin and eosin and examined using light microscope (National model 138, China).

### Pharmacodynamic Model

The pharmacodynamic study was done according to the protocol proposed by Florence et al. [31]. Briefly, male Swiss albino mice weighing from 25 to 35 g were randomly distributed to four different groups (each of 15 mice). Mice were treated with different preparations where group I received intranasal saline as a negative control, group II was administered CZ intranasal solution (CZS<sub>i.n.</sub>), Group III received CZ intravenous solution (CZS<sub>i.v.</sub>) and Group IV was treated by CZ intranasal optimized transferosomes (TF<sub>i.n.</sub>) in a dose of 4mg/kg of body weight [32]. At 15, 30 and 45 minutes after drug administration, five mice were withdrawn from each group and injected intraperitoneal with pentylenetetrazole (PTZ) (100mg/kg of body weight) [33]. The time required for the onset of seizures from the time of injection of PTZ was recorded and taken as an evaluation parameter. Statistical analysis of the obtained results was performed by Post-Hoc test using SPSS 17.0® software (Chicago, USA).

### Results and Discussion

TF are biocompatible and biodegradable as they are made from natural phospholipids similar to liposomes. They have high entrapment efficiency, in case of lipophilic drug near to 90% [34]. Also, TF have the ability to protect the encapsulated drug (such as: protein and peptides) from metabolic degradation. They act as depots, releasing

their contents slowly and gradually. They can be used for both systemic as well as topical delivery of drugs [12]. In the present study, twelve formulae were developed. The ability of TF for direct nose-to-brain delivery was investigated depending on the elasticity properties of the developed nano-carriers.

### In-vitro Evaluation

**Entrapment efficiency:** All of the TF formulations had total drug content  $\approx$  100%. The entrapment efficiency ranged from 25.37% $\pm$ 3.25 (TF10) to 92.31% $\pm$ 0.41 (TF3) (Table 3). EE values were subjected to polynomial analysis using two factor interaction models. The statistical analysis revealed that the three investigated factors, namely, EA type ( $X_1$ ), lipid: EA molar ratio ( $X_2$ ) and initial drug amount ( $X_3$ ) can significantly ( $P < 0.0001$ ) affect the ability of the drug to be incorporated in the TF. Adequate precision was calculated by the Design-Expert® software to demonstrate the signal to noise ratio, whereas a ratio greater than 4 is desirable indicating the validity of the utilized model to navigate the design space [35]. On the other hand, predicted  $R^2$  was calculated as a measure of how good the model could predict a response value by comparing the calculated value with the adjusted  $R^2$  [36].  $R^2$ , adjusted  $R^2$ , predicted  $R^2$  and adequate precision values are presented in Table 4. Adequate precision was 47.145 with reasonable difference between the predicted  $R^2$  (0.9858) and the adjusted  $R^2$  (0.9921).

Formula code*	Y <sub>1</sub> =EE (%)	Y <sub>2</sub> =PS (nm)	Y <sub>3</sub> =PDI	Y <sub>4</sub> =ZP (mV)	Y <sub>5</sub> =DI (g)	Y <sub>6</sub> = Q8h (%)	Y <sub>7</sub> =t50% (h)	Y <sub>8</sub> =DE
TF1	84.22 $\pm$ 1.22	122.5 $\pm$ 2.52	0.24 $\pm$ 0.01	-22.95 $\pm$ 0.07	18.24 $\pm$ 2.17	95.85 $\pm$ 2.73	3.13 $\pm$ 0.62	0.68 $\pm$ 0.00
TF2	86.57 $\pm$ 1.26	138.90 $\pm$ 1.83	0.38 $\pm$ 0.03	-15.95 $\pm$ 0.21	9.20 $\pm$ 1.48	86.22 $\pm$ 0.28	3.55 $\pm$ 0.12	0.68 $\pm$ 0.00
TF3	92.31 $\pm$ 0.41	198.45 $\pm$ 9.12	0.26 $\pm$ 0.02	-15.75 $\pm$ 1.34	4.69 $\pm$ 1.25	85.99 $\pm$ 2.72	6.96 $\pm$ 2.2	0.72 $\pm$ 0.00
TF4	31.40 $\pm$ 0.67	104.50 $\pm$ 0.42	0.54 $\pm$ 0.03	-23.10 $\pm$ 0.42	2.31 $\pm$ 1.75	79.15 $\pm$ 1.08	4.78 $\pm$ 0.03	0.66 $\pm$ 0.02
TF5	41.98 $\pm$ 0.32	116.55 $\pm$ 1.06	0.35 $\pm$ 0.00	-16.00 $\pm$ 0.84	5.61 $\pm$ 1.61	84.28 $\pm$ 0.27	4.34 $\pm$ 0.13	0.65 $\pm$ 0.00
TF6	67.49 $\pm$ 3.48	154.80 $\pm$ 0.98	0.49 $\pm$ 0.04	-19.65 $\pm$ 0.63	4.235 $\pm$ 1.78	85.34 $\pm$ 1.76	12.92 $\pm$ 0.29	0.71 $\pm$ 0.01
TF7	81.84 $\pm$ 0.23	243.80 $\pm$ 12.86	0.36 $\pm$ 0.00	-27.85 $\pm$ 0.91	14.715 $\pm$ 0.47	90.30 $\pm$ 0.43	2.98 $\pm$ 0.01	0.68 $\pm$ 0.02
TF8	83.13 $\pm$ 3.34	858.95 $\pm$ 56.35	0.45 $\pm$ 0.00	-30.85 $\pm$ 0.63	4.71 $\pm$ 1.17	91.61 $\pm$ 0.31	2.60 $\pm$ 0.09	0.66 $\pm$ 0.01



TF9	86.22±0.53	959.70±61.23	0.26±0.00	-27.35±0.91	11.14±1.40	90.50±2.63	2.57±0.31	0.65±0.00
TF10	25.37±3.25	133.75±0.63	0.51±0.10	-19.90±0.42	1.84±1.04	80.79±1.41	3.40±0.11	0.55±0.00
TF11	39.55±0.72	230.90±34.78	0.65±0.03	-30.85±0.49	3.345±1.13	93.42±3.42	2.49±0.10	0.68±0.00
TF12	66.95±0.91	364.50±16.26	0.59±0.28	-21.45±0.49	6.515±1.37	80.12±0.45	7.01±0.31	0.65±0.00

\*Total drug content in all of these formulae was ≈100%.

EE: entrapment efficiency, PS: particle size, PDI: Polydispersity index, ZP: zeta potential Q8h: amount released after 8 hrs and t50%: time for the release of 50% of the drug, DI: deformability index, DE: dissolution efficiency

Table 3: Measured Responses of FFD of Clonazepam Loaded Transferosomes

	R <sup>2</sup> squared	Adjusted R <sup>2</sup>	Predicted R <sup>2</sup>	Adequate precision
EE (%)	0.9952	0.9921	0.9858	47.14
PS (nm)	0.9536	0.9237	0.8635	17.23
PDI	0.7349	0.5645	0.221	5.64
ZP (mV)	0.9516	0.9205	0.8578	18.01
DI (gm)	0.9324	0.8889	0.8012	14.86
Q8h (%)	0.8241	0.7304	0.5498	8.66
t50% (h)	0.9724	0.9547	0.9189	25.34
DE	0.8098	0.6875	0.441	9.752

Table 4: R<sup>2</sup> Squared, Adjusted R<sup>2</sup>, Predicted R<sup>2</sup> and Adequate Precision Values for the Measured Variables According to 22X 31 FFD.

Concerning X<sub>1</sub> (EA type), TF prepared using SDC as a surfactant showed higher entrapment levels in comparison to those prepared using LB. This may be due to the fact that SDC has lower molecular weight if compared to Labrafil, thus smaller weight of surfactant is used in each corresponding ratio. In addition, smaller critical micelle concentration (CMC) of Labrafil in comparison to SDC may indicate the formation of Labrafil micelles hence solubilizing the drug and decreasing the ability of the formed TF to entrap the drug [37].

With respect to X<sub>2</sub> (lipid: edge activator molar ratio), it was obvious that EE increased with decreasing EA molar ratio. High concentration of EA produced a destabilization of the lipid bilayer [38], which could lead to pore formation and thus, decreased EE [39]. The high EA level may also cause solubilization of the drug which in turn may lead to diffusion of the drug into the aqueous medium during preparation of the transferosomes [15]. In addition, upon increasing the EA beyond certain

threshold, micelles are formed, leading to a decrease in EE [40]. These micelles were reported to have a lower drug carrying capacity and poor skin permeation due to their structural features [41].

On the other hand, increasing initial amount of the drug (X<sub>3</sub>) significantly lowered entrapment efficiency. This may be due to the ability of 10mg of CZ to saturate the amount of the lipid used reaching high EE as 92.31%±0.41 (TF3) and 86.22%±0.53 (TF9) indicating that the maximum loading capacity of the transferosomes was reached. Further increase in the nominal drug amount could even decrease the efficiency to entrap the drug [42].

#### Particle size, Polydispersity Index and Zeta Potential:

The mean PS of the prepared TF formulae ranged from 104.5 nm±0.42 (TF4) to 959.7 nm±61.23 (TF9). The statistical analysis using two factor interaction model revealed that the three investigated factors can significantly (P<0.0001) affect PS. Adequate precision was 17.235 with reasonable difference between the predicted R<sup>2</sup> (0.8635) and the adjusted R<sup>2</sup> (0.9237).

In general, TF formulated using SDC as EA showed smaller particle size in comparison to those prepared using LB ( $X_1$ ). This may be due to the higher molecular weight of labrafil, thus the formed vesicle adopted a larger orientation in order to accommodate Labrafil which finally resulted in higher PS [43].

On the other hand, lipid: edge activator molar ratio ( $X_2$ ) significantly affected PS of the prepared transferosomal vesicles where increasing EA molar ratio lead to lower PS. At high level of EA, the interfacial tension will be lowered leading to formation of smaller nanovesicles [44]. Besides, the small vesicle diameters at high EA concentrations might be due to the formation of micelles instead of vesicles which are smaller in size [40]. Surprisingly, statistical analysis revealed that increasing initial drug amount from 10 to 20 mg ( $X_3$ ) lead to higher particle size. This could be mainly explained on the basis that increasing initial drug amount together with lower entrapment efficiency, resulting from this increase, increased the viscosity of the medium which ultimately lead to an increase in the viscous forces resisting droplet breakdown and thus bigger particles are formed, resulting in increased particle size [45].

Concerning PDI, the values obtained ranged between  $0.24 \pm 0.01$  to  $0.65 \pm 0.03$  (Table 3) which could be within the acceptable range according to Cho et al. [46] Who stated that the acceptable range should be between 0.05 to 0.7. Statistical analysis revealed that both  $X_1$  and  $X_3$  could significantly affect the broadness of the size distribution. Generally, LB showed higher PDI values in comparison to SDC. This may be due to the combination of two aforementioned factors, namely, the lower CMC value of Labrafil®- leading to the formation of micelles with small PS- and the larger orientation of the formed transferosomes to accommodate LB molecule. These two factors may result in broader distribution of particle size and thus higher PDI values. On the other hand, increasing initial drug amount from 10 mg to 20 mg ( $X_3$ ) resulted in significantly higher values of PDI. This may be due to more drug molecules precipitation per nucleation site and a less uniform distribution for the PS with higher PDI values [47].

Zeta potential (ZP) can be considered as an important indicator of physical stability of nano-dispersions [48]. A higher electric charge on the surface of the nanoparticles will prevent aggregation because of the strong repellent forces among particles giving more stable dispersions [49]. Generally, ZP values above 20 mV indicate that nano-suspensions are well dispersed with considerable

stability [50]. Results are showed in Table 3. Phospholipon 90G is a zwitterionic compound with an isoelectric point [pI] between 6 and 7. Under experimental conditions of pH around 7, where the pH was higher than its pI, Phospholipon 90G carried a net negative charge. Statistical analysis of ZP revealed that the three investigated factors could significantly influence the measured variable ( $P < 0.0001$ ). Adequate precision was 18.014 with reasonable difference between the predicted  $R^2$  (0.8578) and the adjusted  $R^2$  (0.9205).

Labrafil transferosomes ( $X_1$ ) generally exhibited higher zeta potential. This may be due to the higher amount of labrafil used together with lower CMC value resulting in profound decrease in surface tension and surface free energy at the interface of the formed transferosomal vesicle [51]. This could be also the basis to explain that higher EA molar ratio led to significantly higher ZP values. On increasing EA molar ratio ( $X_2$ ), more surfactant was presented at the surface of the formed TF, thus decreasing the surface tension of the surrounding water layer and decreasing the surface free energy. Knowing that ZP was determined for the filtered transferosomes after removal of the untrapped drug, and that initial drug amount of 10mg showed higher EE if compared to 20mg ( $X_3$ ), could explain the reason why those transferosomes showed higher ( $p < 0.0214$ ) ZP. The anion form of CZ is predominant at neutral pH [52], therefore increasing the EE ( $X_3=10\text{mg}$ ) led to higher ZP levels. In general, negatively charged nano-carriers formulations strongly improve permeation of drugs through biological barriers [53].

**Deformability Index:** The crucial feature which enables transferosomal vesicles to cross biological membranes, when compared with liposomes, is their elasticity [54]. The presence of edge activators (EA) is the reason why transferosomes have the advantage of being ultra-deformable vesicles [55].

The ANOVA of deformability index using two factor model interaction revealed the model to be statistically significant ( $p < 0.0001$ ) with an adequate precision of 14.866 and a reasonable difference between the predicted  $R^2$  (0.8012) and the adjusted  $R^2$  (0.8889). Both  $X_2$  and  $X_3$  were found to have significant effect on the measured variable. For  $X_2$  increasing EA molar ratio was found to increase the ability of the transferosomal vesicle to deform ( $P < 0.0026$ ). These results are in line with those declared by Salama et al. [15] and Gupta et al. [39] stating that increasing EA molar ratio would endly result in

higher deformable properties of the transferosomes. This might be attributed to fluidization of the lipid bilayer.

Interestingly, initial lower drug amount ( $X_3$ ) showed significantly higher deformability index ( $p < 0.0001$ ). This may be due to higher EE recorded with lower initial drug amount, thus increasing the amount of the drug in the formed vesicles. The simultaneous presence of the hydrophobic clonazepam, which is supposed to be intercalated in the bilayer, and EA led to a decrease in lamellar phase stability with a consequent increase in vesicle elasticity [56].

**In-vitro Release:** *In-vitro* cumulative release profiles of the drug from different transferosomes are shown in Figure 1. The passage of CZ from drug solution through the dialysis bag was investigated as control. It reached 82.3% within 2 hrs; this suggested that the drug could freely diffuse through dialysis membrane [57]. By the second hour, the percentage drug released from SDC containing TF (TF1-6) were  $61.57\% \pm 0.74$ ,  $66.22\% \pm 6.46$ ,  $68.49\% \pm 4.57$ ,  $64.40\% \pm 4.45$ ,  $63.19\% \pm 1.51$  and  $67.92\% \pm 0.83$ , respectively. While those prepared using LB (TF7-12) were  $59.92\% \pm 0.78$ ,  $56.68\% \pm 0.84$ ,  $62.49\% \pm 2.12$ ,  $49.81\% \pm 1.04$ ,  $69.34\% \pm 0.16$  and  $68.01\% \pm 0.48$ , respectively.

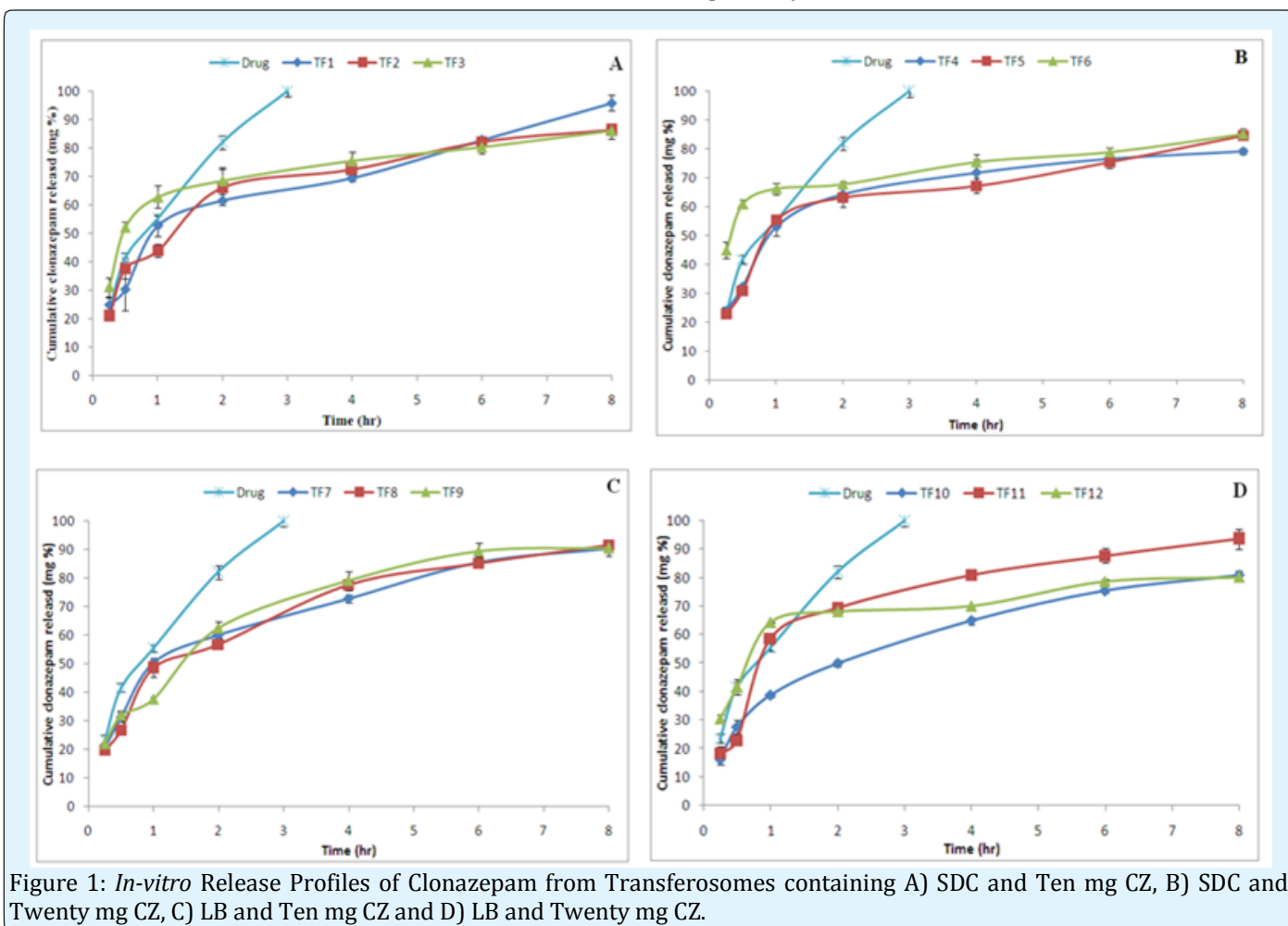


Figure 1: *In-vitro* Release Profiles of Clonazepam from Transferosomes containing A) SDC and Ten mg CZ, B) SDC and Twenty mg CZ, C) LB and Ten mg CZ and D) LB and Twenty mg CZ.

ANOVA of Q8h revealed that increasing initial amount of the drug could significantly ( $P < 0.0001$ ) lead to lower release rate and lower Q8h. This may be due to the fact that lower initial amount of the drug ( $X_3 = 10$  mg) generally showed smaller particle size. Decreased average particle size, increased the effective surface area exposed

to the drug release media, subsequently resulted in an increased initial release [58]. The adequate precision calculated was 7.954 with reasonable difference between the predicted  $R^2$  (0.5498) and the adjusted  $R^2$  (0.7304).

The time required for the release of 50% of the loaded drug ( $t_{50\%}$ ) from transferosomes ranged from  $2.49 \pm 0.1$



(TF11) to  $12.92 \pm 0.29$  (TF6) (Table 3). ANOVA of the obtained results declared that the three investigated factors could significantly affect  $t_{50\%}$  with  $R^2$  value of 0.9724. Adequate precision was 25.34 with reasonable difference between the predicted  $R^2$  (0.9189) and the adjusted  $R^2$  (0.9547). For EA type used ( $X_1$ ), it was deduced that Labrafil® decreased the time required to reach 50% release of the drug ( $t_{50\%}$ ). This may be due to the ability of Labrafil® to destabilize the lipid bilayer [59] leading to faster release of the drug and thus lowering  $t_{50\%}$ .

The effect of the other two variables ( $X_2$  and  $X_3$ ) on  $t_{50\%}$  could be easily explained looking at the particle size analysis. Increasing particle size associated with decreasing EA amount ( $X_2$ ) and increasing initial drug amount ( $X_3$ ) led to lowering total surface area exposed to release medium which in turn decreased the release rate of the drug and increase  $t_{50\%}$ .

Dissolution efficiency for the prepared TF ranged from  $0.55 \pm 0.00$  (TF10) to  $0.72 \pm 0.00$  (TF3) (Table 3). Its ANOVA

points out that all of the three investigated factors had significant effect on the measured variable. SDC ( $X_1$ ) and 10 mg of CZ ( $X_3$ ) generally lead to higher dissolution efficiency in comparison to Labrafil® and 20 mg drug, respectively. This may be due to smaller particle size associated with TF prepared using the formers leading to increased exposed surface area and higher dissolution efficiency. Also TF prepared using 30:1 lipid: EA molar ratio ( $X_2$ ) showed lower DE than those prepared using 10:1. This can be explained in the light that the former had larger particle size in comparison to the latter leading to decreased surface area and increased DE.

The *in-vitro* drug release profiles of the investigated transferosomes could be best fitted to Higuchi-diffusion model (highest  $R^2$ ) (Table 5). These results are in accordance with the release results declared by Patel et al. [60] and Ghanbarzadeh and Arami [61] who formulated curcumin and diclofenac loaded transferosomes, respectively. The release media allows the transferosomal vesicle to swell releasing the drug [62].

Formula code	Coefficient of determination		
	Zero order	First order	Higuchi diffusion
TF1	0.832	0.732	0.941
TF2	0.819	0.678	0.926
TF3	0.758	0.639	0.873
TF4	0.737	0.629	0.871
TF5	0.778	0.645	0.884
TF6	0.784	0.694	0.873
TF7	0.865	0.717	0.956
TF8	0.762	0.623	0.863
TF9	0.868	0.773	0.957
TF10	0.9	0.745	0.978
TF11	0.832	0.732	0.941
TF12	0.669	0.575	0.8

Table 3: Fitting of Clonazepam Release Data from Transferosomes to Certain Models.

**Selection of the optimized TF formula:** In general, the goal of the optimization of pharmaceutical formulations is to determine the levels of variables required to produce a high quality product. Desirability was calculated by Design-Expert® software and considered to optimize the studied responses depending on the provided results. Optimization of all investigated factors simultaneously

was almost impossible as the optimum condition obtained with one response might be associated with the worst result in another one for the same formula.

Clonazepam transferosomes were optimized for the responses  $Y_1$  (EE, %),  $Y_2$  (PS, nm),  $Y_3$  (PDI),  $Y_4$  (ZP, mV),  $Y_5$  (DI, gm),  $Y_6$  (Q8h, %),  $Y_7$  ( $t_{50\%}$ , h) and  $Y_8$  (DE). The aim was to maximize EE, ZP, DI, Q8h and DE while PS, PDI and

t50% were minimized. Highest desirability value obtained was 0.833 and it was devoted to the following composition  $X_1=SDC$   $X_2=10:1$  and  $X_3=10mg$  (TF1).

**Differential Scanning Calorimetry (DSC):** The DSC study was done for CZ, phospholipon 90G, SDC and for CZ loaded transferosomes optimized formula (TF1). Figure 2 shows the DSC thermo gram of CZ with a sharp characteristic endothermic peak at 238 °C indicating its crystalline state. Concerning thermo grams of phospholipon 90G and SDC, small endothermic peaks were detected at 162 °C and 339.7 °C, respectively, indicating their melting points. The DSC thermo gram of CZ optimized formula (TF1) reveals shifting of the SDC endothermic peak to 320 °C and disappearance of the characteristic endothermic peaks of CZ and phospholipon 90G.

Disappearance of the endothermic peak of phospholipid might indicate perturbation of the packing characteristics of the lipid bilayer and its fluidization by SDC [63]. On the other hand, shifting of the SDC endothermic peak might be related to a sort of physical interaction between SDC, phospholipid, and the drug. Disappearance of the characteristic endothermic peak of the drug might indicate the entrapment of the drug in the transferosomes [64].

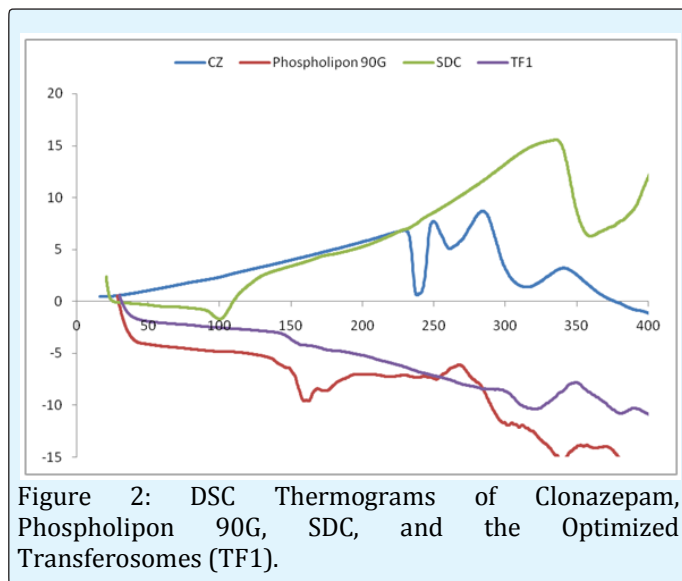


Figure 2: DSC Thermograms of Clonazepam, Phospholipon 90G, SDC, and the Optimized Transferosomes (TF1).

**Transmission Electron Microscopy (TEM):** Photomicrographs of CZ loaded transferosomes (TF1) are illustrated in figure 3 (A and B). It is obvious that the vesicles appeared as irregular spherical shape, which may be due to the presence of higher content of surfactant (3B) [65]. Also, the particle size was in a good agreement with that measured by laser scattering technique.

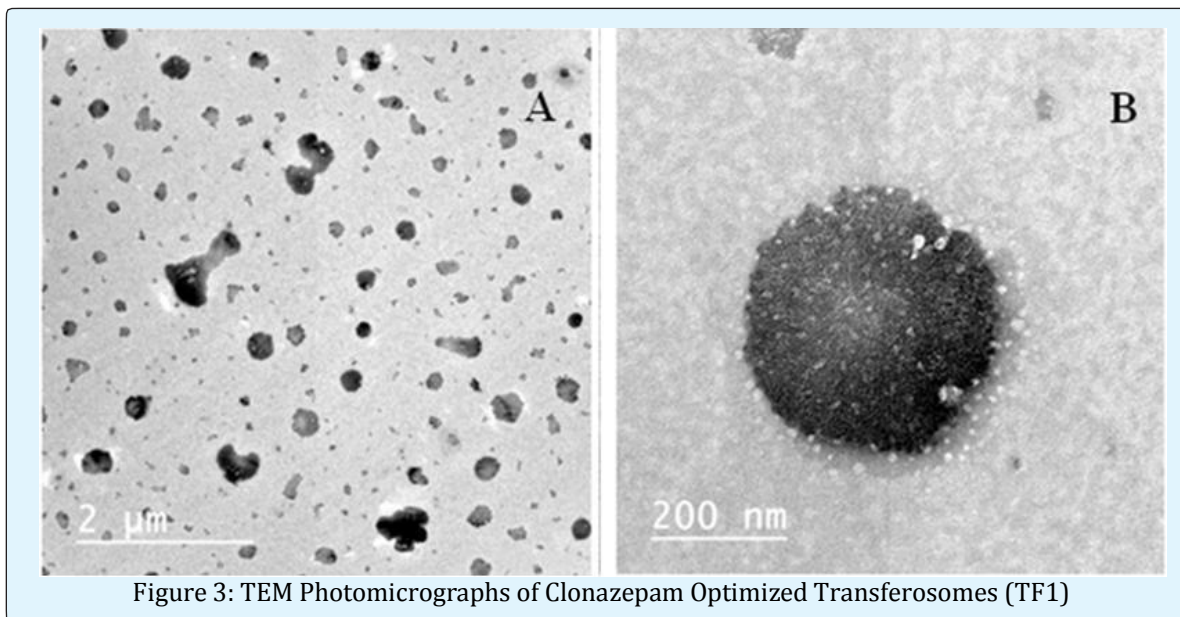


Figure 3: TEM Photomicrographs of Clonazepam Optimized Transferosomes (TF1)

**Effect of storage:** There was no observed aggregation or change in the appearance of the optimized transferosomes (TF1) on storage at 5°C±3 for four weeks.

The recorded EE, PS and Q8h were 83.65 %±4.08, 128.43 nm±2.83 and 92.33 %±0.23, respectively. Statistical analysis revealed that there was no significant difference

( $p > 0.05$ ) in the measured variables of the stored vesicles when compared to the freshly prepared ones.

Figure 4 show that both of the fresh and stored transferosomes exhibited similar release profiles. Calculating similarity factor ( $f_2$ ) produced a value of 72.10 indicating that storage at the specified conditions had no marked effect on the release of the drug. These results are in accordance with Al-mehallawi et al. [37] who formulated transferosomes containing the drug ciprofloxacin. Their results demonstrated physical stability of the developed nanocarriers to six month at the same storage conditions.

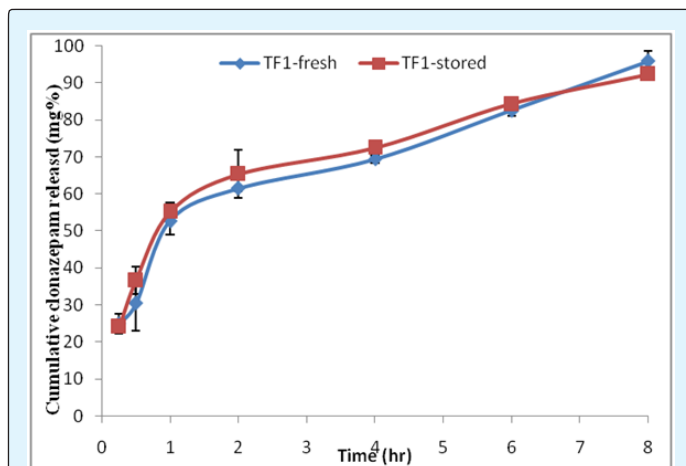


Figure 4: *In-vitro* Release Profiles of Clonazepam from Optimized Transferosomes TF1 (fresh and stored).

#### **Ex-vivo Assessment of Nasal Cytotoxicity**

As depicted in (Figure: 5A), anterior part of the nasal mucosa treated with PBS as a negative control, revealed no change in the histological structures with normal stratified squamousepithelium and intact underlying connective tissue containing sebaceous glands and hair follicles. Upon exposure to isopropyl alcohol, (figure: 5B) sloughing of the epidermal lining with disfiguration of the underlying tissue was observed. On the other hand, applying TF1 to the anterior segment of the sheep nasal mucosa led to only some focal thinning of the epithelium without affecting the underlying tissue.

Examining the posterior part (Figure: 5C), treated with pH 6.4 PBS, revealed normal pseudo stratified columnar epithelium with submucosa, submucosal glands and cartilaginous layer (Figure: 6A). On exposure to isopropyl alcohol, sloughing of the epithelium was noticed with complete distortion of the submucosal layer (Figure: 6B).

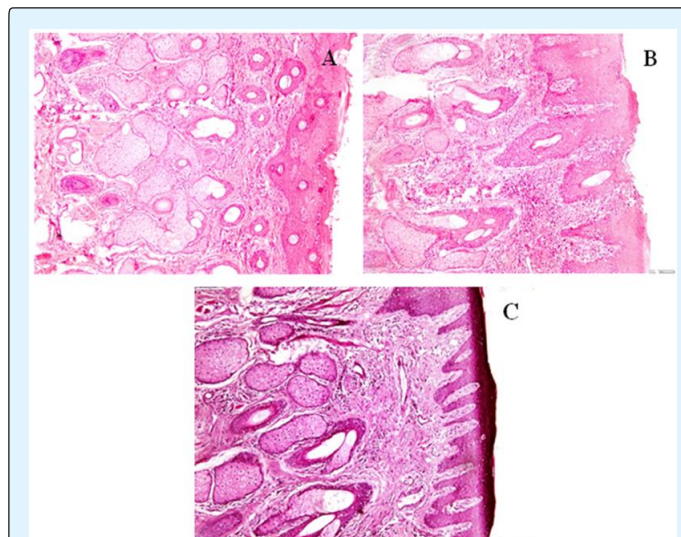


Figure 5: Photomicrographs of the Anterior Segments of Sheep Nasal Mucosa Treated with pH 6.4 PBS (negative control, A), Isopropyl alcohol (positive control, B), and CZ-Loaded Transferosomes (C) (100X).

For TF1 (Figure: 6C), minor inflammation was noticed in the posterior part of the sheep nasal mucosa associated with number of inflammatory cells infiltration. These results are in line with Salama et al. [15] who developed olanzapine transferosomes prepared using SDC. Upon application of the formed TF, mild-to-moderate reversible inflammation of the nasal epithelium was reported.

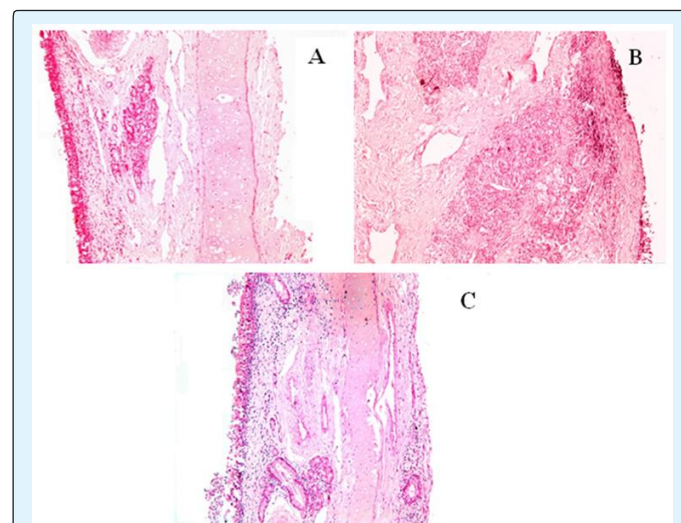


Figure 6: Photomicrographs of the Posterior Segments of Sheep Nasal Mucosa Treated with pH 6.4 PBS (negative control, A), Isopropyl alcohol (positive control, B) and CZ-Loaded Transferosomes (C) (100X).



## Pharmacodynamic Studies

The ability of different CZ formulation administered intranasally and intravenously to protect mice against PTZ-induced seizures was measured adopting the protocol developed by Florence et al. [31]. PTZ was administered intraperitoneal after 15, 30, 45 minutes post treatment with CZ formulations. Saline was taken as negative control. The time required for elucidation of convulsions after PTZ injection was determined and taken as a measure for the ability of the different formulation to protect the mice. The results are presented in Table 6. The time recorded in group I receiving intranasal saline for elucidation of seizures was  $59.33 \pm 3.6$ ,  $56 \pm 6.02$  and

$59.66 \pm 2.08$  sec in 15, 30 and 45 min groups, respectively. Although intranasal CZ solution group showed prolongation in the onset of seizures to  $98 \pm 10.5$ ,  $84.33 \pm 10.53$  and  $71.33 \pm 9.07$  sec for 15, 30 and 45 min groups, respectively, statistical analysis revealed these values to be insignificant. This may be due to the limited ability of the intranasal (I.N.) solution to deliver the drug in adequate concentration to the brain. On the other hand, CZ solution administered intravenously could offer protection and significant prolongation ( $p < 0.05$ ) in the groups receiving CZ 30 and 45 min before PTZ injection. However, it didn't show any significant protection for 15 min treatment group in comparison to control group.

Formula Administered	Administration after 15 min (Mean±S.D)	Administration after 30 min (Mean±S.D)	Administration after 45 min (Mean±S.D)
Normal saline	$59.33 \pm 3.6$	$56.00 \pm 6.02$	$59.66 \pm 2.08$
CZS <sub>(i.n.)</sub>	$98.00 \pm 10.5$	$84.33 \pm 10.53$	$71.33 \pm 9.07$
CZS <sub>(i.v.)</sub>	$62.00 \pm 6.55$	$160.6 \pm 7.02$	$139.00 \pm 8.18$
TF1 <sub>(i.n.)</sub>	$207.00 \pm 33.95$	$250.00 \pm 56.04$	$243.33 \pm 39.50$

SD: standard deviation, (n = 5), CZS: clonazepam solution, i.v: intravenous, i.n: intranasal.

Table 4: Time (sec) for the Development of Seizures in Male Wister Mice

Furthermore, the time for elucidation of seizures in group IV treated with TF1 intranasally was significantly ( $P < 0.0001$ ) higher than all treatment groups and the control group at all time intervals. This confirms the ability of the developed transferosomes to directly deliver the drug to the brain circumventing BBB and also declares the aptitude of the formula to allow drug accumulation in high concentration in the brain improving its efficacy and decreasing its side effects.

## Conclusion

In this study, incorporation of CZ in ultradeformable lipid containing vesicles, transferosomes, was successfully done adopting TFH technique. The effect of different formulation variables was studied using full factorial design. The ability of the lipid carrier to entrap the drug was confirmed by TEM and DSC studies. Furthermore, the efficacy of the optimized transferosomes (TF1) to directly deliver the drug in adequate concentration to the brain was evidenced by PTZ seizures challenge. Thus, TF1 (formulated using 10mg CZ, 15.6 mg SDC and 284.4 mg Phospholipon 90G ) could be considered as an excellent optimized for immediate control of status epilepticus.

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