

# Reorientation of Hydrocarbons Chains in Gasoline Fuels under Exposure to Static Magnetic Field: A Starting Point for Increasing Energy Efficiency of Motor Vehicles

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## Research Article

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## Abstract

We report the result that the reorientation of hydrocarbons chain in gasoline fuel is induced by an applied static magnetic field. Indeed, 1 h 30 min exposure of samples of gasoline fuel to a 150 mT static magnetic field provided the result that CH<sub>2</sub> bending vibration around 1465 cm<sup>-1</sup>, CH<sub>3</sub> symmetrical band at 1378 cm<sup>-1</sup> and C–C stretching at 1610 cm<sup>-1</sup> decreased significantly, whereas CH<sub>2</sub> twisting band around 1230 cm<sup>-1</sup> increased significantly after exposure. These findings demonstrated that a reorientation of hydrocarbons chains occurs under an applied magnetic field. This relevant result can be used to increase the energy efficiency of motor vehicles.

**Keywords:** Gasoline fuel; Hydrocarbons chains; Static magnetic field; Diamagnetism; FTIR spectroscopy

## Introduction

In this review paper the response of gasoline fuel under exposure to a static magnetic field (SMF) studied by Fourier Transform Infrared (FTIR) spectroscopy analysis was reported in order to demonstrate that changes of vibration bands can induce an increase of the combustion of air-fuel mixture. Gasoline fuel is obtained from petroleum crude oil and is composed by a mixture of

hydrocarbons, mostly alkanes, i.e. on the form C<sub>n</sub> H<sub>2n+2</sub>. The primary constituent of fuel gasoline is n-butane (C<sub>4</sub>H<sub>10</sub>), n-pentane (C<sub>5</sub>H<sub>12</sub>), n-hexane (C<sub>6</sub>H<sub>14</sub>), n-heptane (C<sub>7</sub>H<sub>16</sub>) and above all n-octane (C<sub>8</sub>H<sub>18</sub>). The properties of gasoline are dependent on the species contained in the blend [1,2].

We know that hydrocarbons in motors gasoline fuel burn mixing with air by means of the combustion

represented by the reaction with oxygen, whose the most representative is the chemical reaction with octane, which produces heat and pressure within the cylinder during the four-stroke combustion cycle [3,4]. A relevant problem in motor vehicles is how maximize the energy efficiency of this combustion cycle. To this aim, the engineers should choose the correct air/fuel ratio.

The power output of motor vehicles strictly depends on the amount of fuel that can be combusted in the cylinders. Nevertheless, only about 20% of the total energy obtained by the combustion process can be used, because the remaining 80% is lost to friction and wasted as heat. Scientists, government agencies and vehicle manufacturers have largely searched methods to increase the efficiency of motor vehicles. Indeed, the potential to improve fuel efficiency by gasoline's combustion in engines is enormous [5,6]. Such research has been encouraged by government agencies also to reduce the emissions limiting the output and reactivity of pollutants, in particular the emission of CO<sub>2</sub> [7].

Recent studies showed that natural magnets located in a point around the tube where gasoline flows towards carburetor can increase gasoline's combustion efficiency and the output energy due to the enhancing of reaction with the oxygen during the combustion process [8-11]. This result was not confirmed by experiments coordinated by the Environmental Protection Agency (EPA) in USA, whose result showed that no significant difference are detected between the use of magnetic field devices inserted in the fuel plant of motor vehicles and the traditional fuel plant in analogue vehicles [12].

Nevertheless, we think that applying a SMF to the entire gasoline tank in motor vehicles would produce different effects from the traditional case as the whole amount of fuel gasoline would be subject to the SMF, contrary to previous studies in which only a small area of the tube where gasoline flows was subjected to a magnetic field. In order to demonstrate this assumption, the response of gasoline vibrations to SMF was studied by means of FTIR spectroscopy. This technique has been successfully used in previous studies to highlight the characteristics of petroleum compounds [13-15].

## Materials and Methods

### Gasoline Samples and Experimental Set-Up

Different gasoline samples were collected from various commercial processing plants and subjected to the following assay. The octane number of gasoline was 95. Each sample consisted of 15 ml of gasoline placed in small

glass containers. Either exposed and control samples were located in the same room at the temperature at 20°C.

Exposed samples were placed between two Helmholtz coils, at the center of the coils distance, that were driven by a DC generator producing a uniform magnetic field intensity at 150 mT, following the theory of Helmholtz coils, as accurately described in [16,17].

### Infrared Spectroscopy

FTIR spectroscopy was applied to gasoline samples by using a spectrometer Vertex 80v of Bruker Optics. Gasoline samples of 200 µl were placed between a pair of CaF<sub>2</sub> windows and for each spectrum 64 interferograms were collected with a spectral resolution of 4 cm<sup>-1</sup>. Interactive baseline correction, smoothing correction, vector area normalization were used for exposed and control samples as accurately described in [18-20]

Finally, statistical analysis was applied to 18 different samples of gasoline fuel using Student's t-test for comparisons between exposed and unexposed samples, with p-values less than 0.05 considered significant.

## Results and Discussion

Representative transmittance spectra of exposed and unexposed samples after 15 min exposure in the region 3000-1200 cm<sup>-1</sup> were reported in Figure 1, in which exposed and unexposed samples spectra are represented by red and blue color, respectively.

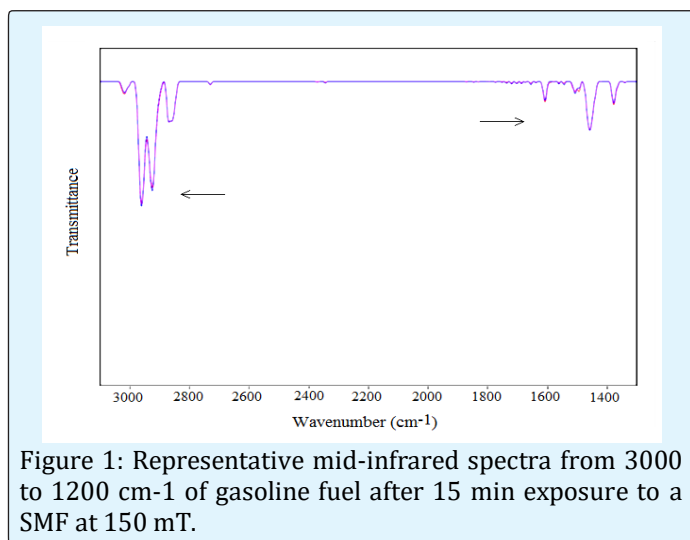


Figure 1: Representative mid-infrared spectra from 3000 to 1200 cm<sup>-1</sup> of gasoline fuel after 15 min exposure to a SMF at 150 mT.

The asymmetric stretching vibrations of methyl <sup>as</sup>CH<sub>3</sub> and methylene as CH<sub>2</sub> groups can be observed at 2963 cm<sup>-1</sup> and 2922 cm<sup>-1</sup>, respectively; also, the symmetric

stretching of methyl  $^s\text{CH}_3$  and methylene  $^s\text{CH}_2$  are represented by the vibrations at 2885 and 2840  $\text{cm}^{-1}$ , respectively [21,22]. No appreciable changes in these vibrations were observed under exposure to SMF at 150 mT up to 1 h 30 min. This result was confirmed applying Fourier Self-deconvolution (FSD) analysis [23].

Furthermore, the band around 1465  $\text{cm}^{-1}$  observed in the spectra, assigned at  $\text{CH}_2$  scissoring vibration did not change significantly after 15 min exposure (Figure 2A), but decreased significantly ( $p < 0.05$ ) after 1 h 30 min of exposure (Figure 2B), showing that exposure to SMF can change  $\text{CH}_2$  bending vibration in gasoline fuel [24-26].

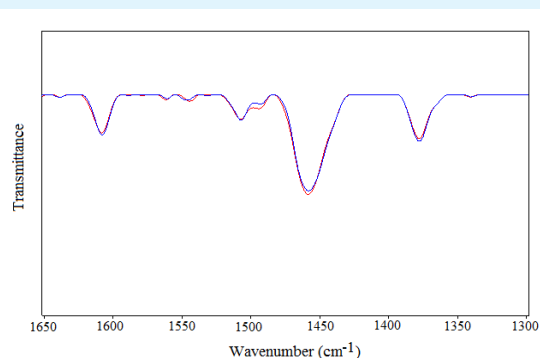


Figure 2A: Representative FTIR spectra from 1650 to 1300  $\text{cm}^{-1}$  of gasoline fuel after 15 min of exposure to a static magnetic field at the intensity of 150 mT. The spectra of exposed samples are represented in red color.

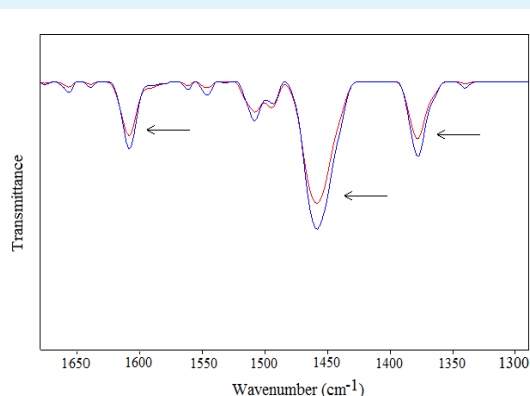


Figure 2B: Representative FTIR spectra from 1650 to 1300  $\text{cm}^{-1}$  of gasoline fuel after 1 h 30 min of exposure to a static magnetic field at the intensity of 150 mT.  $\text{CH}_2$  bending vibrations at 1610, 1465 and 1378  $\text{cm}^{-1}$  (pointed out by arrows) decreased in intensity after exposure. The spectra of exposed samples are represented in red color.

In addition, the peak at 1378  $\text{cm}^{-1}$  can be attributed to  $\text{CH}_3$  symmetrical deformation band and the bands observed at 1655, 1610 and 1522  $\text{cm}^{-1}$  can be attributed

to C-C stretching vibrations [29]. The C-C aromatic ring stretch is represented by the vibration at 1495  $\text{cm}^{-1}$ . Also, the strong vibrations at 1610 and 1378  $\text{cm}^{-1}$  decreased in intensity significantly ( $p < 0.05$ ) after exposure up to 1 h 30 min (Figure 2B).

The classical theory of diamagnetism is able to explain the change in intensity of the main vibration bands of gasoline fuel, observed after exposure to SMF. Indeed, aliphatic hydrocarbons that compose gasoline fuel are diamagnetic substances with diamagnetic susceptibility whose intensity increases with increasing of the number  $n$  of carbon atoms in the chain  $\text{C}_n\text{H}_{2n+2}$  [30]. Diamagnetism consists of a magnetic field created in diamagnetic materials which opposes to the external applied magnetic field. Thus, a SMF applied to gasoline fuel should induce a magnetic field in the fuel which opposes to the applied field.

As a result, the application of a SMF on gasoline samples should induce the alignment of gasoline chains with their axes parallel to the field, opposing to it. This phenomenon was already observed in macromolecules like polymers, that align along the direction of an applied SMF which can induce the reorientation of the polyethylene chains towards the direction of the field [31-33]. In this scenario, the decrease in intensity of bending vibrations observed after exposure to SMF can be explained as follows.

The  $\text{CH}_2$  chains of octane molecule are at equilibrium with a nonzero dipole moment. Under the exposure to SMF, these chains begin their motion aligning with the applied field and opposing to it, causing that the angle bends become larger than in the absence of a magnetic field (Figure 3A & Figure 3B). As a result, the dipole moment is reduced so that corresponding bending vibrations of  $\text{CH}_2$  and  $\text{CH}_3$  decrease in intensity because the reorientation of hydrocarbons chains.

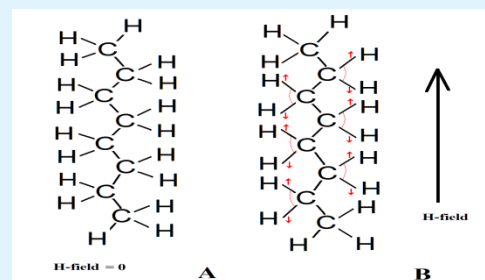


Figure 3A: Scheme of the octane chemical structure in gasoline fuel before exposure to a SMF.

Figure 3B: Scheme of the octane chemical structure in gasoline fuel representing its rearrangement in plane after exposure to a SMF.

Also, the increase in intensity of the vibration at  $1230\text{ cm}^{-1}$  which was observed by Calabrò and Magazù, can find its explanation. In fact, this band can be attributed to out of plane twisting of  $\text{CH}_2$  group. The deviation of direction of C-H linkages from  $\text{CH}_2$  plane should induce an increase of dipole moment which opposes to the external field.

The enlargement of the angles in plane and out of plane between the C-H linkages of  $\text{CH}_2$  group should induce an increasing of combustion process. Indeed, the application of a SMF should induce opposite spinning electrons to have parallel spins, generating a magnetic field opposing to the external field [34]. As a result, parallel spinning electrons could react with oxygen atoms more rapidly than molecules with paired electrons spinning opposite directions, so that the combustion process increases with increasing of the applied SMF. Also previous study are in agreement with our result. Indeed, it was observed that hydrocarbons viscosity decreases with increasing of the applied SMF, so that better atomization of the fuel should verify [35]. In view of these findings, it can be hypothesized to plan a gasoline tank embedded in the magnetic field produced by a permanent electromagnet, driven by the same electric plant which is in the motor vehicle, so that the magnetic field originated by the magnet can increase during the motion of the vehicle [36-39].

## Conclusions

Hydrocarbons in gasoline fuel under exposure to a SMF at the intensity of 150 mT were studied using FTIR spectroscopy. First, no appreciable change of symmetric and asymmetric stretching of  $\text{CH}_3$  and  $\text{CH}_2$  vibration bands was observed after exposure to SMF. In contrast, bending vibration bands decreased significantly after 1 h 30 min of exposure such as vibration band at  $1465\text{ cm}^{-1}$  assigned at  $\text{CH}_2$  scissoring vibration. This result can be explained by the theory of diamagnetism, assuming that hydrocarbons chains reoriented towards the direction of the applied SMF, opposing to it. Also the methyl symmetrical deformation band at  $1378\text{ cm}^{-1}$  and C-C stretching vibrations at  $1610\text{ cm}^{-1}$  decreased in intensity after exposure to SMF, confirming this scenario. These findings showed that an enlargement of the angles in plane and out of plane between the C-H linkages of  $\text{CH}_2$  group occurred after exposure to SMF, favoring the combustion with oxygen. As a result, the energy efficiency of motor vehicles could increase hypothesizing to plan a gasoline tank embedded in the magnetic field produced by a permanent electromagnet, driven by the same electric plant which is in the motor vehicle.

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